

**Table S.1.-** Details of the crystal data, structural resolution and refinement procedure for Ni<sub>3</sub>(C<sub>4</sub>H<sub>4</sub>N<sub>2</sub>)<sub>3</sub>(V<sub>8</sub>O<sub>23</sub>).

Formula	C6 H6 N3 Ni1.5 O11.5 V4
Molecular weight, g·mol <sup>-1</sup>	595.96
Crystalline system	Triclinic
Space group (N <sup>o</sup> .)	P-1 (2)
a, Å	8.556(2)
b, Å	9.117(2)
c, Å	12.619(3)
α, °	71.05(2)
β, °	83.48(2)
γ, °	61.32(3)
V, Å <sup>3</sup>	815.7(3)
Z	2
ρ <sub>obs</sub> , ρ <sub>calc</sub> , gr·cm <sup>-3</sup>	2.41(1), 2.426
F(000)	578
<b>Collection data</b>	
Temperature, K	293(2)
Diffractionmeter	Oxford Diffraction Xcalibur2
μ, mm <sup>-1</sup>	3.962
Radiation, λ(Mo Kα), Å	0.71073
Cristal size, mm	0.12 x 0.06 x 0.04
Range θ, °	2.87-26.37
Interval h, k, l	-10 ≤ h ≤ 8, -11 ≤ k ≤ 10, -15 ≤ l ≤ 15
N <sup>o</sup> of measured refelctions	6134
N <sup>o</sup> of independent reflections	3269
R (int)	0.0376
<b>Refinement</b>	
Data/Restrictions/Parameter	3269/0/238
R factors [I>2σ(I)]	R1=0.0272, wR2=0.0409
R factors [all data]	R1=0.0460, wR2=0.0439
Máx. y min. of electronic residual density, e.Å <sup>-3</sup>	0.411, -0.359
G. O. F	0.901

$$R1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|} ; wR2 = \sqrt{\frac{\sum w(|F_o| - |F_c|)^2}{\sum w|F_o|^2}}$$

$$w = 1/[\sigma^2|F_o|^2 + (xp)^2 + yp] \text{ with } p = [\max|F_o|^2 + 2|F_c|^2]/3, \text{ where } x = 0.0045, y = 0$$

**Table S.2.-** Selected bond distances (Å) and angles (°) for Ni<sub>3</sub>(C<sub>4</sub>H<sub>4</sub>N<sub>2</sub>)<sub>3</sub>(V<sub>8</sub>O<sub>23</sub>).**Bond distances**

<i>Ni(1)O<sub>4</sub>N<sub>2</sub> octahedron</i>		<i>Ni(2)O<sub>4</sub>N<sub>2</sub> octahedron</i>		<i>V(1)O<sub>4</sub> tetrahedron</i>	
N1-O(7) <sup>i</sup>	2.028(2)	Ni2-O(11)	2.034(2)	V1-O(2) <sup>iv</sup>	1.648(2)
Ni1-O(7) <sup>ii</sup>	2.028(2)	Ni2-O(4)	2.058(2)	V1-O(4)	1.659(2)
Ni1-O(6) <sup>i</sup>	2.054(2)	Ni2-O(2)	2.069(2)	V1-O(3)	1.800(2)
Ni1-O(6) <sup>ii</sup>	2.054(2)	Ni2-N(1)	2.080(2)	V1-O(1)	1.831(2)
Ni1-N(2)	2.095(2)	Ni2-O(10)	2.091(2)		
Ni1-N(2) <sup>iii</sup>	2.095(2)	Ni2-N(3)	2.106(2)		
<i>V(2)O<sub>4</sub> tetrahedron</i>		<i>V(3)O<sub>4</sub> tetrahedron</i>		<i>V(4)O<sub>4</sub> tetrahedron</i>	
V2-O(10)	1.649(2)	V3-O(7)	1.643(2)	V4-O(9)	1.599(2)
V2-O(6)	1.663(2)	V3-O(11)	1.658(2)	V4-O(5)	1.770(2)
V2-O(3) <sup>iv</sup>	1.790(2)	V3-O(12)	1.7833(9)	V4-O(1) <sup>v</sup>	1.773(2)
V2-O(5)	1.819(2)	V3-O(8)	1.821(2)	V4-O(8) <sup>vi</sup>	1.781(2)
<i>(C<sub>4</sub>H<sub>4</sub>N<sub>2</sub>) pyrazine</i>					
N(1)-C(1)	1.337(4)	N(1)-C(4)	1.351(4)	N(2)-C(3)	1.338(4)
N(2)-C(2)	1.349(4)	N(3)-C(5)	1.332(4)	N(3)-C(6) <sup>x</sup>	1.347(4)
C(1)-C(2)	1.393(4)	C(5)-C(6)	1.387(4)	C(6)-N(3) <sup>x</sup>	1.347(4)

**Bond angles**

<i>Ni(1)O<sub>4</sub>N<sub>2</sub> octahedron</i>		<i>Ni(2)O<sub>4</sub>N<sub>2</sub> octahedron</i>	
O(7) <sup>i</sup> -Ni1-O(7) <sup>ii</sup>	180.0(0)	O(11)-Ni2-O(4)	178.49(8)
O(7) <sup>i</sup> -Ni1-O(6) <sup>i</sup>	90.14(9)	O(11)-Ni2-O(4)	90.29(8)
O(7) <sup>ii</sup> -Ni1-O(6) <sup>i</sup>	89.86(9)	O(4)-Ni2-O(2)	89.50(8)
O(7) <sup>i</sup> -Ni1-O(6) <sup>ii</sup>	89.86(9)	O(11)-Ni2-N(1)	91.91(9)
O(7) <sup>ii</sup> -Ni1-O(6) <sup>ii</sup>	90.14(9)	O(4)-Ni2-N(1)	88.43(9)
O(6) <sup>i</sup> -Ni1-O(6) <sup>ii</sup>	180.0(0)	O(2)-Ni2-N(1)	174.95(9)
O(7) <sup>i</sup> -Ni1-N(2)	99.14(9)	O(11)-Ni2-O(10)	89.26(8)
O(7) <sup>ii</sup> -Ni1-N(2)	90.86(9)	O(4)-Ni2-O(10)	92.23(8)
O(6) <sup>i</sup> -Ni1-N(2)	88.71(9)	O(2)-Ni2-O(10)	89.48(8)
O(6) <sup>ii</sup> -Ni1-N(2)	91.29(9)	N(1)-Ni2-O(10)	86.00(9)
O(7) <sup>i</sup> -Ni1-N(2) <sup>iii</sup>	90.86(9)	O(11)-Ni2-N(3)	89.34(9)
O(7) <sup>ii</sup> -Ni1-N(2) <sup>iii</sup>	89.14(9)	O(4)-Ni2-N(3)	89.16(9)
O(6) <sup>i</sup> -Ni1-N(2) <sup>iii</sup>	91.29(9)	O(2)-Ni2-N(3)	86.02(9)
O(6) <sup>ii</sup> -Ni1-N(2) <sup>iii</sup>	88.71(9)	N(1)-Ni2-N(3)	98.55(9)
N(2)-Ni1-N(2) <sup>iii</sup>	180.0(0)	O(10)-Ni2-N(3)	175.28(9)

<i>V(1)O<sub>4</sub> tetrahedron</i>		<i>V(2)O<sub>4</sub> tetrahedron</i>	
O(2) <sup>iv</sup> -V1-O(4)	111.5(1)	O(10)-V2-O(6)	111.3(1)
O(2) <sup>iv</sup> -V1-O(3)	107.6(1)	O(10)-V2-O(3) <sup>iv</sup>	107.8(1)
O(4)-V1-O(3)	106.1(1)	O(6)-V2-O(3) <sup>iv</sup>	107.2(1)
O(2) <sup>iv</sup> -V1-O(1)	108.8(1)	O(10)-V2-O(5)	109.9(1)
O(4)-V1-O(1)	114.1(1)	O(6)-V2-O(5)	108.4(1)
O(3)-V1-O(1)	108.41(9)	O(3) <sup>iv</sup> -V2-O(5)	112.39(9)

<i>V(3)O<sub>4</sub> tetrahedron</i>		<i>V(4)O<sub>4</sub> tetrahedron</i>	
O(7)-V3-O(11)	109.3(1)	O(9)-V4-O(5)	110.2(1)
O(7)-V3-O(12)	110.05(8)	O(9)-V4-O(1) <sup>v</sup>	109.4(1)
O(11)-V3-O(12)	109.71(8)	O(5)-V4-O(1) <sup>v</sup>	108.5(1)
O(7)-V3-O(8)	105.4(1)	O(9)-V4-O(8) <sup>vi</sup>	107.7(2)
O(11)-V3-O(8)	108.2(1)	O(5)-V4-O(8) <sup>vi</sup>	109.1(1)
O(12)-V3-O(8)	113.96(8)	O(1) <sup>v</sup> -V4-O(8) <sup>vi</sup>	112.0(1)

*(C<sub>4</sub>H<sub>4</sub>N<sub>2</sub>) pyrazine*

C(1)-N1-C(4)	117.1(3)
C(1)-N1-Ni(2)	118.4(2)
C(4)-N1-Ni(2)	124.4(2)
C(3)-N2-C(2)	116.9(3)
C(3)-N2-Ni(1)	122.7(2)
C(2)-N2-Ni(1)	120.5(2)
C(5)-N3-C(6) <sup>x</sup>	116.1(2)
C(5)-N3-Ni(2)	119.7(2)
C(6) <sup>x</sup> -N3-Ni(2)	123.6(2)
N(1)-C1-C(2)	121.8(3)
N(2)-C2-C(1)	120.9(3)
N(2)-C3-C(4)	122.4(3)
N(1)-C4-C(3)	120.9(3)
N(3)-C5-C(6)	123.0(3)
N(3) <sup>x</sup> -C6-C(5)	120.9(3)

*Symmetry codes :*

i= -x+2,-y+1,-z	ii= x,y+1,z	iii= -x+2,-y+2,-z	iv= -x+2,-y,-z+1
v= -x+3,-y,-z+1	vi= x+1,y,z	vii=x,y-1,-z+1	viii= x-1,-y+1,-z
ix= -x+1,-y+1,-z	x= -x+1,-y+1,-z+1		