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Data Article

Geochemical dataset of high-pressure acid migmatites from the Cabo Ortegal Complex (NW Spain)



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ABSTRACT

This brief note presents geochemical data from rock samples from the Cabo Ortegal Complex (NW Spain). The samples belong to acid lithologies within the mainly basic to intermediate granulite unit that have been poorly investigated so far. For this communication, five samples of the migmatites and an amphibolitic enclave within them have been analysed. The whole-rock major and trace-element analyses were accomplished by means of Q-ICPMS. The dataset provides new and useful information relevant to the origin of the acid migmatites and can be used in addition to information from neighbouring lithologies to enhance understanding of the geological evolution of the Western Variscan Belt.

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Specifications Table

Subject	Earth Sciences
Specific subject area	Geochemistry and Petrology
Type of data	Table
51	Figure
How data were acquired	Analysis of whole-rock composition (major and trace elements) using quadrupole ICPMS (Thermo Fisher XSeries II)
Data format	Raw and Analyzed
Parameters for data collection	The samples were collected from the poorly studied acid lithologies within the extensively investigated, mainly basic to intermediate in composition, granulite unit of the Cabo Ortegal Complex.
Description of data collection	The description of the data collection is presented in the Experimental Design, Materials. and Methods section.
Data source location	Samples storage at:
	University of the Basque Country UPV/EHU, Science and Technology Faculty,
	SGIker research facility, Leioa (Spain)
	Samples collected at the Cabo Ortegal Complex, Bacariza Formation (NW
	Iberia). See coordinates in [1].
Data accessibility	With the article
Related research article	A. Beranoaguirre, S. Garcia de Madinabeitia, M. E. Sanchez-Lorda, P. Puelles, B.
	Abalos, J. I. Gil Ibarguchi, U-Pb, Hf isotope and REE constraints on
	high-pressure acid migmatites from the Cabo Ortegal Complex (NW Spain):
	New evidence of short-duration metamorphism in a Variscan subduction
	channel, Lithos 372-373 (2020) 105,660. doi:10.1016/j.lithos.2020.105660

Value of the Data

- The data provide geochemical information to classify and characterize the acid migmatites of the granulite unit at Cabo Ortegal.
- The data can be used to better understand the petrogenesis of the rocks.
- The data can be re-used, together with additional information, to gain further insights regarding the origin and subsequent evolution of the high-pressure metamorphic nappes of the Cabo Ortegal Complex.
- The data presented in this article complement the data obtained from the previous study [1].

1. Data Description

Geochemical data from poorly known acid migmatites exposed in the Cabo Ortegal Complex (Western Variscides, NW Spain) are presented here. The migmatites, previously called granulitic orthogneisses [2], are part of the high-pressure granulite unit, tectonically placed between the eclogite and ultramafic (peridotite-pyroxenite) units [3]. Except for the so-called felsic gneiss sample and the amphibolitic enclave, all the other samples bear evidence of partial melting. Detailed petrographic information of those migmatites can be found elsewhere [1,2]. Due to the ubiquity of the mafic granulites, earlier geochemical works in the area have been focused on those lithotypes [e.g. 4], leaving aside the scarce acid migmatite outcrops. This paper intends to fill this gap on geochemical data and for that purpose, six samples from the biotite-absent muscovite bearing migmatites (samples T-4 and CO-35a), muscovite- and biotite-bearing migmatites (CO-55a), biotite-bearing migmatites lacking muscovite (CO-5b, including an amphibolitic enclave), and mica-lacking felsic gneisses (CO-49) have been studied. Accompanying frequent minerals include garnet, quartz, feldspars and rutile, and, in some cases zoisite/clinozoisite, kyanite and antiperthitic plagioclase.

Table 1 shows the results of the whole rock major and trace element analyses of the selected samples, together with two samples from the literature [4] used in Fig. 1 (see below) for comparative purposes. All samples analysed are quite evolved, with SiO_2 contents varying from 50.2%

 Table 1

 Major and trace element analyses of acid migmatites from the Cabo Ortegal Complex. * from [4].

Sample	CO-08-5b	$CO-08-5b_{Enc}$	CO-35a	CO-49	CO-55a	T-4	AMP-463*	AMP-466*
Major elements (%)								
SiO ₂	58.06	52.17	56.22	59.68	50.21	71.83	43.60	46.10
TiO ₂	1.31	1.26	1.02	0.58	1.69	0.48	1.96	3.90
Al ₂ O ₃	17.72	16.88	18.67	15.93	20.19	13.12	15.20	13.80
Fe_2O_3	9.18	10.98	9.73	9.50	12.86	5.12	18.20	15.80
MnO	0.17	0.12	0.16	0.19	0.24	0.11	0.24	0.28
MgO	3.78	6.67	3.08	2.34	4.57	1.10	6.5	6.22
CaO	3.56	6.56	4.47	7.81	4.46	2.57	10.9	10.8
Na ₂ O	3.15	2.34	3.04	3.28	3.66	1.99	1.23	0.99
K ₂ O	1.85	1.46	3.11	3.28	3.66	2.52	0.18	0.34
P ₂ O ₅	0.21	0.13	0.11	0.10	0.12	0.12	0.05	0.47
LOI	0.58	0.98	0.66	0.11	0.76	0.82	0.00	0.50
Total	99.57	99.55	100.27	102.79	102.41	99.78	98.06	99.20
Trace elements (ppm)								
Sc	23.5	25.9	4.53	4.84	4.15	24.9	-	-
v	139.1	228.1	171.6	124.5	268.9	35	-	-
Cr	85.3	280.0	158.6	204.9	211.9	54.3	200	300
Со	19.0	96.1	34.77	19.61	70.61	75.2	55	36
Ni	47.0	82.4	46.97	b.d.l.	218.2	6.1	b.d.l.	55
Cu	45.1	46.2	21.9	15.64	21.01	b.d.l.	-	-
Zn	54.6	125.4	56.92	55.92	91.11	38.1	-	-
Rb	54.0	32.9	109.8	7,782	69.42	54.4	4	7
Sr	197.9	120.6	201.7	280.1	265.9	151	153	62
Y	35.1	35.8	35.08	19.77	68.37	100	b.d.l.	46
Zr	194.1	124.4	231.3	57.58	380.9	401.2	37	235
Nb	9.2	6.1	13.23	b.d.l.	25.29	10	5	7
Ba	488.7	258.7	780.3	75.72	491.8	915	167	252
La	20.5	15.0	38.07	9.229	86.56	111	2	10.7
Ce	55.6	33.7	95.3	17.43	176.5	229	5	28
Pr	5.9	4.2	11.46	2.631	23.65	29.4	-	-
Nd	22.5	18.0	44 11	11.61	93 56	114	4	21
Sm	52	54	9 179	3 059	18.8	23.3	117	6.61
Eu	12	15	2 157	1 316	3 473	2.7	0.69	2.66
Gd	52	6.4	11 28	3 668	23 34	19.5	-	-
Th	0.9	11	1 372	0.605	2 727	3.03	03	17
Dv	61	6.6	7525	3 636	14 79	18 5	19	10
Ho	10	11	1498	0 779	3 041	3 11	-	-
Fr	34	3.5	4 465	2 322	917	10.3	_	-
Tm	0.6	0.6	0.652	0.366	1367	171	-	-
Vh	3.6	3.5	4 4 2 8	2 535	9.031	10.1	124	5 75
Iu	0.6	0.6	0.628	0 388	174	1 79	0.19	0.86
Hf	5.7	39	6 574	1 699	10.76	12 75	1	7
Тэ	0.8	10	1 0 0 1	0 194	1962	12.75	bdl	, hdl
Th	0.0	3.0	8 723	1/05	18 15	283	b.d.i.	b.d.i.
III II	3.1 1 2	10	0.725	1.405	2 2 4 2	20.5	0.0.1.	0.4
U	1.0	1.0	2.10	1.101	2.242	2.4	0.2	0.4

up to 71.8% and MgO <5%, except for the amphibolite nodule in sample CO-5b. Their freshness is confirmed by the < 1% values for loss on ignition.

Harker diagrams for major element oxides are presented in Fig. 1. As a comparison, we also plotted the geochemical data of two already published data of mafic and rare ultramafic (pyrigarnite) granulites within the same unit (from [4]). All major elements except K_2O exhibit a coherent decreasing linear trend with respect to increasing SiO₂, above all observable in MgO and Fe₂O₃ diagrams. This is consistent with a differentiation process for magmatic protoliths of the granulite unit. The main difference between mafic and acid rocks is related to the higher alkali (Na₂O and K₂O) contents as well as distinctively lower Fe₂O₃ and CaO components in the latter. Indeed, the amount of K₂O in acid lithotypes is responsible for the characteristic white mica growth in the migmatites. Regarding the trace elements, the most significant feature is the



Fig. 1. Plots of SiO_2 content against different major elements oxide (Harker diagram) of the samples. Black dots correspond to the samples analysed here, white dots are from [4]. See text for further explanations.

increasing of U content increasing SiO₂. It is also worth mentioning the low Zr and Hf content on the mica-lacking gneisses (sample CO-49), where lower amount of zircon was also observed. Likewise, the gneisses have a distinctive lower content of REE in comparison to the amount observed in the white mica rich lithotypes ($\Sigma REE < 60 \ \mu g/g$ in the gneisses against $\Sigma REE >$ 550 $\mu g/g$ in the biotite-absent muscovite bearing migmatite, sample T-4).

According to the classification of [5] (Fig. 2a), the migmatite samples are peraluminous, except for the amphibolite enclave in sample CO-5b that plots in the metaluminous field. When plotting in the ternary AFM diagram of [6] (Fig. 2b) the samples show a calc-alkaline affinity but close to the boundary with the tholeiitic series. The rocks studied here do not plot in any particular field of the ones proposed by [7] (Fig. 2c), pointing to a progressive trend. Similarly, the tectonic discrimination diagrams of [8] are also inconclusive (Fig. 2d), with cases straddling the boundaries between volcanic-arc and within-plate granites.

Trace element concentrations (Table 1) are reported in primitive mantle-normalized multielement diagrams and chondrite-normalized REE patterns (Fig. 3, normalized to [9]). They display patterns typical for volcanic-arc scenarios, with enrichments in large-ion lithophile elements (Rb, Ba) and depletion in Nb and Ta (Fig. 3a), as well as enrichment in LREE (La, Ce, Pr) over HREE (Fig. 3b). A notable feature of these patterns is their negative anomaly in Sr and Eu in the partially melted samples, i.e. all except the felsic gneiss, which is slightly enriched in Sr and Eu. This reflects the more evolved nature of the samples because Sr becomes relatively depleted with respect to its neighbouring elements during plagioclase fractionation. Indeed, the



Fig. 2. Samples of the acid migmatites plotted in classification diagrams proposed by (a) [5], (b) [6], (c) [7], and (d) [8].

positive anomaly in Sr and Eu observed in the felsic gneisses may indicate the cumulative nature of this lithology.

2. Experimental Design, Materials and Methods

The sample processing was accomplished at the sample preparation laboratories of the Geology Department at the University of the Basque Country UPV/EHU. 5 to 10 kgs of the sample (depending on grain size) were ground using first a jaw crusher (Retsch bb200 Wolframcarbid) and subsequently a Retsch RS1 type ring-mill with widia rings.

The measurement protocol was mostly based on the procedures of [10], and it was as follows: 250 mg of sample and 500 mg of LiBO₂ were put into the Pt(Au) crucible with three to four drops of LiBr solution as a non-wetting agent. The mixture was fused and poured onto a polypropylene beaker containing 100 ml of diluted HNO3:HF acid mixture. The solution was stirred ca. 10 min to ensure total dissolution. The samples were consequently analysed using a Thermo Fisher XSeriesII quadrupole ICPMS at the SGIker Research Facilities of the University of the Basque Country UPV/EHU. The precision for all analytes is in general < 2%. The LOI was determined by heating the samples using a muffle furnace with a temperature of 1000 °C. Before and after heating the samples, their masses were measured using a balance weighing with the accuracy ± 0.0002 g so the weight loss was used to calculate the LOI.



Fig. 3. Multi-element and REE patterns of the studied samples. Normalization values are the 'pyrolite silicate Earth' for the primitive mantle, and C1 chondrites [9].

Ethics Statement

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Not applicable.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships which have, or could be perceived to have, influenced the work reported in this article.

CRediT Author Statement

Aratz Beranoaguirre: Investigation, Formal analysis, Data curation, Writing – original draft; **Sonia Garcia de Madinabeitia:** Formal analysis, Writing – original draft; **Maria Eugenia Sanchez-Lorda:** Formal analysis, Writing – original draft; **Pablo Puelles:** Data curation, Writing – original draft; **Benito Abalos:** Data curation, Writing – original draft; **Jose Ignacio Gil-Ibarguchi:** Data curation, Writing – original draft.

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