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Elemental imaging approach to assess the ability of subaerial biofilms growing on constructions located in tropical climates as potential biomonitors of atmospheric heavy metals pollution

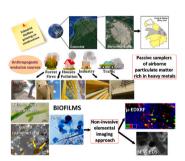
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HIGHLIGHTS

- Subaerial biofilms (SABs) for passive biomonitoring studies of air quality in tropical cities.
- Deposition of metallic PM in SABs growing naturally on buildings from tropical cities.
- µ-EDXRF and SEM-EDS single point and imaging in SABs to assessment air pollution sources.
- Analytical methodologies for fast and cost-effective assessment of PM deposited in SABs.

G R A P H I C A L A B S T R A C T



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ABSTRACT

Over the last decades, the concern about air pollution has increased significantly, especially in urban areas. Active sampling of air pollutants requires specific instrumentation not always available in all the laboratories. Passive sampling has a lower cost than active alternatives but still requires efforts to cover extensive areas. The use of biological systems as passive samplers might be a solution that provides information about air pollution to assist decision-makers in environmental health and urban planning. This study aims to employ subaerial biofilms (SABs) growing naturally on façades of historical and recent constructions as natural passive biomonitors of atmospheric heavy metals pollution. Concretely, SABs spontaneously growing on constructions located in a tropical climate, like the one of the city of Barranquilla (Colombia), have been used to develop the

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Scanning Electron Microscopy coupled with Energy Dispersive X-ray spectrometry methodological approach here presented as an alternative to SABS grown under laboratory conditions. After a proper identification of the biocolonizers in the SAB through taxonomic and morphological observations, the study of the particulate matter accumulated on the SABs of five constructions was conducted under a multianalytical approach based mainly on elemental imaging studies by micro Energy Dispersive X-ray fluorescence spectrometry (μ -EDXRF) and Scanning Electron Microscopy coupled with Energy Dispersive X-ray spectrometry (SEM-EDS) techniques, trying to reduce the time needed and associated costs. This methodology allowed to discriminate metals that are part of the original structure of the SABs, from those coming from the anthropogenic emissions. The whole methodology applied assisted the identification of the main metallic particles that could be associated with nearby anthropogenic sources of emission such as Zn, Fe, Mn, Ni and Ti by SEM-EDS and by μ -EDXRF Ba, Sb, Sn, Cl and Br apart others; revealing that it could be used as a good alternative for a rapid screening of the atmospheric heavy metals pollution.

1. Introduction

Atmospheric pollutants have been the subject of interest by the scientific community due to their negative impact on human health (Karri et al., 2016), food security (Sun et al., 2017), and terrestrial and aquatic ecosystems (Szynkowska et al., 2018; Tchounwou et al., 2012). The official reports of the World Health Organization indicate that, in different regions of the world, the concentration of particulate matter (PM) harmful to the human health has been increasing in the last years (World Health Organization, 2016). In urban areas, some studies correlate the high amount of PM rich in heavy metals with specific health problems (Combes and Franchineau, 2019; Glencross et al., 2020; Schraufnagel et al., 2019; Sun et al., 2020). Atmospheric particulate matter rich in heavy metals is usually originated from anthropogenic activities (i.e., road traffic, greenhouse effect, industrial activities, thermal power plants, etc.) and natural events (i.e., volcanic eruptions, forest fires, organic matter decomposition, etc.) (Lelieveld et al., 2015; Haikerwal et al., 2015; Zhang and Batterman, 2013; Querol et al., 2007).

Many are the works dealing with the study of atmospheric airborne particulate matter deposition on leaves, trees, lichens, mosses, soils and sediments (Castaneda-Miranda et al., 2020; Roy et al., 2020; Contardo et al., 2020; Boquete et al., 2020; Islam and Saikia, 2020; Chiaia--Hernández et al., 2020), but scarce those which evaluate its impact on building materials (Sesana et al., 2018). Since decades, most of the studies have focused on the evaluation of the influence of acid gases on the state of conservation of construction materials (Grossi and Brimblecombe, 2002; Prieto-Taboada et al., 2013). One of the main pathologies studies since years in historical and contemporary constructions is the formation of black crusts (Di Turo et al., 2016; Morillas et al., 2016; Martínez-Arkarazo et al., 2007; Saiz-Jimenez, 1993). These crusts are usually formed by the sulfation of the calcium carbonate present on the construction material mainly by the action of atmospheric SO₂ (Ruffolo et al., 2015). The final grey/black color of this kind of crusts is associated with the accumulation of airborne particulate matter on its porous structure. All this has motivated some scholars to use this kind of crusts as passive samplers of airborne particulate matter (Morillas et al., 2016) or even as markers of environmental changes (García-Florentino et al.,

Similarly, biofilms show a great affinity to metallic contaminants and other polar or positively charged molecules due to the presence of charged sites in the heteropolymers of the extracellular matrix as a result of the ionization of functional groups, such as carboxyl and amino groups, sulfates, uronic acids, pentoses and deoxysugars (Cheng et al., 2018; Rossi and De Philippis, 2015). In constructions, under adequate environmental conditions of humidity, sunlight, temperature and pH of the substrate, the colonizers frequently form consortia of phototrophic microorganisms (algae and cyanobacteria) and heterotrophs (bacteria, fungi and protozoa) at the solid/air interface that are called subaerial biofilms (SAB) (Gulotta et al., 2018). Phototrophic microorganisms have been considered primary colonizers, conditioning the nature of building material to subsequent grow heterotrophic microorganisms such as fungi (Villa and Cappitelli, 2019; Crispim et al., 2003). Usually, biodeterioration processes involve many groups of microorganisms that

co-exist simultaneously in the susbtrate, forming structured, heterogeneous, and genetically diverse communities protected by a 3D exopolysaccharides matrix (Rossi and De Philippis, 2015; Lembre et al., 2012). This last matrix is primarily responsible for the structural and functional integrity of the SAB (Crispim and Gaylarde, 2005) and, at the same time, determines its physicochemical properties. In functional terms, biofilm matrices play an important role in the biosorption of various contaminants, such as heavy metal ions in aquatic and terrestrial environments (Kurniawan and Fukuda, 2022).

SABs can be also formed spontaneously on the surface of constructions and they do not necessarily imply a "biodeteriorative" role as is frequently thought. In this sense, different scholars have reported the neutral or even bioprotective effect of SABs on stones and other constructing materials under certain environmental conditions (Ramírez et al., 2010; Villa et al., 2015, 2016; Zerboni et al., 2022). In these cases, SABs can act as a barrier that blocks the interaction between the construction materials with the atmosphere. This physical barrier prevents the transformation of the original mineralogical composition due to the interaction with other environmental deterioration agents (e.g., acid gases).

Commonly, to evaluate the nature and concentration of the heavy metals in the atmospheric particulate matter, cascade impactors are usually employed (Chang et al., 2000; Botle et al., 2020; Liu et al., 2022; Zhang et al., 2018). This kind of instruments, thanks to the pumping of the air, collect the particles on filters. Some authors have reported the use of lichens and mosses as natural biomonitors or passive samplers of atmospheric heavy metal pollution (Ogunkunle et al., 2016; Salo et al., 2012; Chakrabortty and Paratkar, 2006; Gallego-Cartagena et al., 2021; Lin, 2015; Wannaz et al., 2013). A recent work confirmed the accumulation of heavy metals associated to airborne PM on algae biolfims (García-Florentino et al., 2018a).

Considering these experimental evidences, the use of SABs that grow spontaneously on constructions deserve to be evaluated as a potential cost-effective passive alternative to extract a general and rapid panoramic view of the possible impact of heavy metals-rich airborne PM in urban areas. Appart from the PM sampling procedure, the development of a fast and cost-effective analytical methodology would highly contribute to the whole strategy. This approach could be useful to assist stakeholders in environmental health and urban planning to make quick decisions about the status of air pollution. Moreover, these analytical methodologies could be also useful in the future to assess the ability of green façades including specific biological organisms (e.g., mosses and algae) in the filtering of atmospheric particulate matter (Srbinovska et al., 2021; Paull et al., 2020; Radić et al., 2019).

In this work, to test the ability of SABs retaining atmospheric PM rich in heavy metals, those formed on the façades of constructions from an urban city, such as the one of Barranquilla (Colombia), were considered. To evaluate their role as retainers of atmospheric PM, first, the taxonomic identification of the biocolonizers in the SABs was conducted. Once elementally and mineralogical characterized the plasters/mortars of the façades where the SABs were samples, the airborne PM trapped on them was studied. To this aim, a non-invasive analytical strategy based on elemental imaging analysis, which does not require any sample

preparation has been designed. To evaluate the distribution and abundance of heavy metals present in the particles accumulated in the SABs Scanning Electron Microscopy coupled with Energy Dispersive X-ray spectrometry (SEM-EDS) and micro-Energy Dispersive X-ray Fluorescence spectrometry (μ -EDXRF) was used. The obtained experimental evidences were employed to extract conclusions about the impact of nearby anthropogenic sources contributing to the emission of airborne PM rich in heavy metals.

2. Materials and methods

2.1. Sampling locations, climatic conditions and anthropogenic pollution sources

In order to prevent from possible washing of the PM trapped on the SABs by the action of rain, sampling was carried out during the dry season (December–January) (Zhou et al., 2022; Mariani and de Mello, 2007; Rubio et al., 2001). The samples were collected from the north facing façade, the preferential orientation where the SABs have been formed (Häubner et al., 2006). Concretely, samples from five constructions from the North Historic Center of the Special, Industrial, and Port District of Barranquilla, capital of Atlantic Department (Colombia) were obtained (Fig. 1). In Barranquilla, the tropical savannah climate predominates according to the Köppen-Geiger climate classification with an average annual temperature of 27.1 °C, presenting minimum temperatures of 25.2 °C in February and maximum of 29.4 °C in July. Two moderate wind flows prevail with northeast (42.7%) and north

(25%) directions, respectively. Frequencies relative to the East, Southeast and South are also presented, with 5.8%, 6.1% and 6.1% of observations, respectively (IDEAM, 1997). The average annual rainfall is 814 mm, where October reaches the highest rainfall peak with 1396 mm and February the lowest peak with 22 mm. The percentage of annual relative humidity (% RH) is 80%, exhibiting the maximum peak in October with 85.10% and the minimum in February with 78.21%. The mean annual solar luminosity is around 3406.8 h (Agudelo-Castañeda et al., 2020; Ramírez-Cerpa et al., 2017). In addition to the climatic factors of the city described, other environmental factors of an anthropogenic nature coexist that assist the growth of biofilms on the surface of the selected buildings, such as humidity induced by air conditioning devices, irrigation of vertical gardens on terraces and the discharge of water through river gutters (Table S1). The façades of the selected constructions showed signs of aesthetic alterations and deterioration related to biofilms (Fig. S1 until S5).

To select the studied constructions, those areas from the locality with identified sources of anthropic activities that could contribute to the emission of particulate matter rich in heavy metals were considered. Generally speaking, the Historic Center of Barranquilla also suffer the diffuse impact of the recurrent forest fires that usually occur in the Vía Isla de Salamanca Park during the dry season (Nolte, 2016).

The first sampling point (B1) was the building of the *Universidad del Atlántico* headquarters, located in the city's downtown area (Fig. S1). This sampling point is located 50 m from a highway with high road traffic, 100 m from auto shops, and 200 m from fuel (Fig. S1A). The second sampling point (B2) was the *Catedral Metropolitana de*

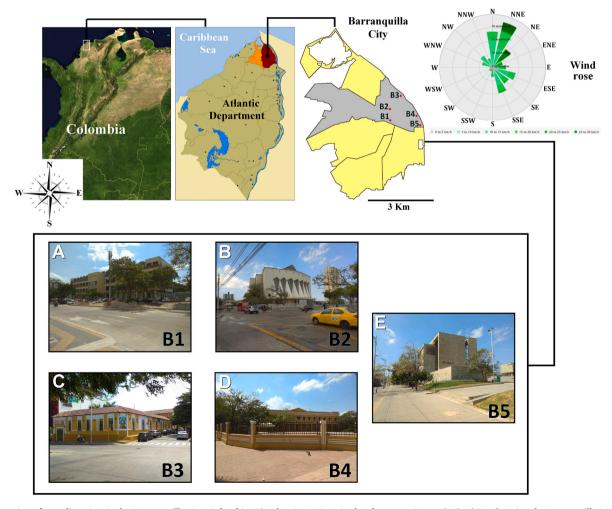


Fig. 1. Location of sampling sites in the Barranquilla city, Colombia. A) Atlantico University headquarters Center (B1); B) María Reina de Barranquilla Metropolitan Cathedral (B2), C) Casa del Carnaval (B3), D) Caribbean Pilot Library (B4) and E) Caribbean Museum (B5).

Barranguilla, located in the city's downtown area (Fig. S2). This sampling point is at a distance of 50 m from an urban road with high traffic, 100 m from an urban bus station and 500 m from the city's industrial complex. The third sampling point (B3) was the historical construction of the Casa del Carnaval (Fig. S3). This sampling point is located 300 m from a food and paper processing industry, 50 m from a highway with high vehicular traffic. The fourth sampling point (B4) was the historical construction of the Biblioteca Piloto del Caribe (Fig. S4). The building is located 500 m from the city's industrial complex. Among the main sources of contamination are the stock centers for the recycling of metallic materials (100-150 m away) roads with high traffic of mediumload transport vehicles (50 m away) and the presence of a traffic light (70 m away). Finally, the fifth sampling point (B5) was the Museo del Caribe building (Fig. S5). This sampling point is located 50 m from an urban bus station and a road with high vehicular traffic, 100 m from shopping centers with high pedestrian traffic, and 200 m from a recycling stock center for metallic materials.

2.2. SABs sampling and description

Considering the historic value of the constructions and in order to minimize the damage during samplings, three small fragments of 4–9 x 4–9 mm² from the areas colonized by the SABs from each construction respectively were taken. The samples were randomly extracted from the first 2 m heigh of the façade considering the floor as zero. To characterize the elemental and mineralogical composition of the mortars/substrates where SABs have been formed, also three mortar fragment of less than $10\times10~{\rm mm}^2$, from each construction respectively, were randomly taken from areas not affected by the SABs but close to them. For the sampling of the areas colonized (SABs) sterile scalpel blades were used, cleaning from one to each sampling. For the mortar fragments extraction, a hammer and a chisel were used. The samples of construction material colonized with biofilms were stored in plastic microtubes. The samples of construction material without colonization were stored in Ziploc bags and preserved in Petri dishes.

For the growth and subsequent taxonomic characterization of eukaryotic and prokaryotic algae, biofilm scrapings were conducted on the SABs and the scrapped material was introduced into Eppendorf vials enriched with Guillard's medium f/2 and BG-11. Such vials were placed in a culture chamber for 12 day at a temperature of 28 °C \pm 2 °C, pH 7.5 \pm 1.2 and a photoperiod of 8 h of light and 16 h of darkness, respectively. After this time, samples of algae and cyanobacteria developed with a Pasteur micropipette were taken from the culture vials and placed on slides for observations under an optical microscope, and taxonomic identification was carried out through morphological descriptions and implementation of taxonomic keys suggested by different scholars (Anagnostidis and Komárek, 1988; Anagnostidis, 1986; López-Bautista et al., 2007; Prescott, 1964).

In the case of SAB samples intended for the identification of heterotrophic microorganisms (fungi), biofilm scraping samples were inoculated in triplicate in Petri dishes enriched with Sabouraud Agar Glucose and Chloramphenicol Glucose Agar culture media (Merck, Germany). The inocula were incubated in a model E23 oven (BINDER, Germany) at 30 °C for 5 days. After this time, each sample was subjected to the Ridell technique to facilitate the selection and handling of the developed colonies (Weber and Pitt, 2000). The microscopic fungal samples were separated from the colonies with clamps and placed on a slide with a drop of lactophenol or tween 20. For morphological descriptions, microscopic observations the structures (for example, type of mycelia, presence of septa, etc) with 10X and 40X objectives were made. In the case of observing spore ornamentation and color, a 100X immersion objective was used. The interpretation and analysis of the morphological characteristics and taxonomic identification were carried out according to the studies presented elsewhere (Bartnicki-Garcia, 1968; Kidd et al., 2016).

2.3. Instrumentation

The mineralogical characterization of the mortars by means of X-ray diffraction (XRD) previously required the pulverizing of the samples by manual grinding in an agate mortar. For the XRD analyses, 0.5-1.0 g were used per sample. One sample was generated from each of the three mortar samples obtained from each building. The samples were idependently analyzed showing the representative results obtained from the three of them. The analyses were conducted using a powder diffractometer PANalytical Xpert PRO instrument equipped with a copper tube ($\lambda_{\text{CuK}\alpha\text{media}} = 1.5418 \text{ Å}, \lambda_{\text{CuK}\alpha 1} = 1.54060 \text{ Å}, \lambda_{\text{CuK}\alpha 2} =$ 1.54439 Å), a vertical goniometer (Bragg-Brentano geometry), a programmable divergence aperture, an automatic interchange of samples, a secondary monochromator of graphite and a PixCel detector. The measurement conditions were 40 kV and 40 mA, with an angular range (20) scanned between 5 and 70°. Moreover, X'pert HighScore (PANalytical) software combined with the specific powder diffraction file database (International Centre for Diffraction Data-ICDD, Pennsylvania, USA) was used for the data treatment of the diffractograms and the identification of the present mineral phases.

The elemental analysis of the SABs and their corresponding mortar was conducted using the EDXRF spectrometer M4 TORNADO (Bruker Nano GmbH, Berlin, Germany). With this instrument, multi-point and imaging analyses were conducted. For the quantitative elemental analysis of the mortars, one pressed pellet was prepared from each of the three samples. A multi-point measurement strategy was used. Fifty points randomly distributed were programmed in each pellet at a lateral resolution of 1 mm. Measurements were acquired during 300 s using an Rh tube operating at 50 kV and 700 μ A. The quantitative elemental information was obtained using previously optimized empirical calibrations described elsewhere (García-Florentino et al., 2018b). For certain elements (Co, V, Cr, Ni, Cu, Br, As and Ba) semi-quantitative information was only possible to obtain. Results were calculated as the average of the results obtained from the three analyzed pellets from each sampling location.

To study the SABs, multi point and imaging analyses were conducted. For the imaging study, a microfocus side window Rh tube powered by a low-power H.V. generator and cooled by air was used. Thanks to the implemented poly-capillary lens, measurements were acquired at down to 25 μm of lateral resolution. The spot size varies as a function of the energy, 17 μm at 2.3 keV and 32 μm at 18.3 keV. For spectral acquisition, the X-ray tube was programmed at its maximum voltage and current of 50 kV and 600 μA, respectively. The detection of the fluorescence radiation was performed by an XFlash silicon drift detector with 30 mm² sensitive areas and an energy resolution of 145 eV for Mn-K α . To improve the detection of the lightest elements (Z < 11), filters were not used, and measurements were acquired under vacuum (20 mba) using the diaphragm pump MV 10 N VARIO-B. For the focus of the area under study, two video microscopes were used to explore the sample under low magnification (1 cm² area), and the other one to perform the final focusing (1 mm² area). The elemental distribution maps were obtained after the deconvolution of the sum spectrum of the whole mapped area, and they were shown according to the K_{α} line of each element except for Pb, for which the L_{α} line was used. The acquisition and processing of the individual spectra and mappings were conducted using the ESPRIT family software (Bruker Nano GmbH, Berlin, Germany). In this work, the most representative distribution maps are shown. Selected samples of SABs were scrapped from the mortar acting as substrate, and they were analyzed following the multipoint strategy previously described for the mortar samples. In this case, only semi-quantitative results were obtained.

Elemental distribution maps obtained through μ -EDXRF were also used to compare, which are the heavy metals present at higher concentrations in each SAB obtained from each construction. For that, the concentration (semi-quantitative values) of each heavy metal detected was divided by the area of studied SAB sample. As three samples were

maped from each constructions, the results were offered in $\mu g/g/mm^2$ together with the standard deviation. For the comparative study, as the constructions from which the samples B1, B2 and B3 are coetaneous, we assume that SABs would have formed in a similar period of time. In contrast, the construction from which B5 samples were taken, was recently erected. This last has been taken into account in the comparison of the numerical values obtained.

To determine the composition of the individual particles and aggregates accumulated in the SABs, SEM-EDS analyses were also conducted. The selected samples were mounted on an aluminum pin stub fixed to a carbon tape. The three SAB samples from each location were visualized under the SEM in order to select the one with a higher number of particles deposited as the representative and considered one for the analyses. For the image acquisition, the surfaces of SABs were coated with gold to ensure their conductivity and, therefore, to obtain microscopic images of good quality. For the SEM-EDS analyses, an EVO®40 Scanning Electron Microscope (Carl Zeiss NTS GmbH, Germany) coupled to an XMax Energy-Dispersive X-ray spectrometer (Oxford Instruments, Abingdon, Oxfordshire, United Kingdom) was used. SEM images were obtained at a low vacuum employing an acceleration voltage of 30 kV and a 10-400 µm working distance for the image acquisitions and elemental analysis (single point and imaging). Different magnifications (reaching up to 6800X) were used for secondary electron images, and an integration time of 50 s was employed to improve the signal-to-noise ratio. The EDS spectra were acquired and treated using the INCA suite software (version 4.13) (Oxford Instruments, Abingdon, Oxfordshire, United Kingdom).

3. Results and discussion

3.1. Identification of the microorganisms present in the SABs

Phototrophic microorganisms at sampled locations showed a light green to dark green or grayish patchy distribution on the building surface. These surfaces were often affected by humidity due to water retained within the substrates, splashing of water by rain gutters, or garden irrigation activities (Table S1 and Fig. S1 until Fig. S5).

The SAB samples analyzed in this study are compact filamentous

matrix, with a regular shape totally distributed on the surface of the mortars (Fig. 2a). Exceptions are the samples B4 (Fig. 2c) and B5 (Fig. 2b). The first one represents a black non-filamentous compact matrix and the second one shows a non-compact and diffuse structure.

Microscopic, morphological, and taxonomic analysis of the microorganisms isolated from the SABs indicated the presence of ten phototrophic microorganisms and four type of fungi (Table 1). In this study, it was possible to identify eight bacteria from Phylum Cyanobacteria in each analyzed SAB sample. The identified cyanobacteria genus were Phormidium sp., Oscillatoria sp., Chroococcus sp., Gloeocapsa sp., Cyanosarcina sp., Leptolyngbya sp., Lyngbya sp. and Asterocapsa sp. Additionally, in the B5 sample, Chlorella sp. and Trentepohlia sp. from Phylum Chlorophyta were also identified. In the majority of the SAB samples, fungal species such as Aspergillus niger, Aspergillus fumigatus, Penicillium sp., and Fusarium sp. from Phylum Ascomycota were identified.

Observations of the phototrophic microorganisms collected from the biofilms under the light microscope revealed the presence of filamentous forms of *Phormidium* sp., *Oscillatoria* sp., *Leptolyngbya* sp. and *Lyngbya* sp. (Fig. S6) while under canopy the coccoid cyanobacteria *Chroococcus* sp., *Cyanosarcina* sp., *Asterocapsa* sp. and *Gloeocapsa* sp. were observed in most of the samples (Fig. S7).

Fungal activity is significantly higher in tropical climates (high temperatures and humidity), such as the one of Barranquilla (Pinto et al., 2019). Concretely, in the SABs under study an extended presence of fungi from Phylum Ascomycota with a higher-prevalence of the species *A. niger*, *A. fumigatus*, and the genera *Penicillium* sp. and *Fusarium* sp was observed. In the SABs, (Fig. S8), these microorganisms successfully growth due to their synergistic interaction with cyanobacteria (Gaylarde, 2020; Scheerer et al., 2009; Cutler and Viles, 2010).

Sample B5 can be considered a specific case in which *Chlorella* sp. and *Trentepohlia* sp. algae were identified as accessory species among the abundant composition in the SAB of filamentous (*Phormidium* sp., *Oscillatoria* sp. and *Lyngbya* sp.) and coccoid cyanobacteria *Gloeocapsa* sp. The presence of cyanobacteria with hyaline or gelatinous envelopes likely favors the fixation to the substrate of the identified fungal and phototrophic microorganisms. Experimental studies indicate that *Chroococcus* sp. has a reasonably thick mucus layer that promotes the maintenance of Chlorophyta agglomerations such as *Chlorella* sp. and

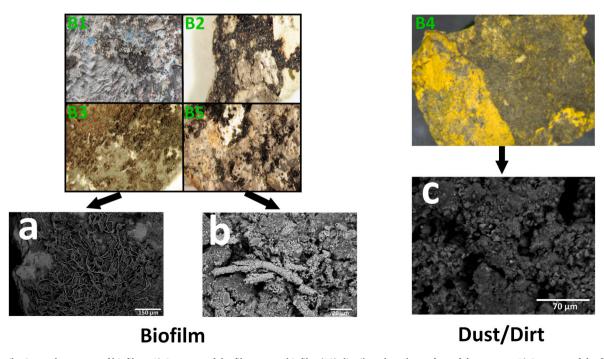


Fig. 2. Distribution and structure of biofilms. A) Structure of the filamentous biofilm (B3) distributed on the surface of the mortars. B) Structure of the filamentous biofilm (B5) with diffuse distribution and not very compact. C) Non-filamentous compact matrix with regular distribution on the surface of the substrate (B4).

Table 1Phototrophic microorganisms and fungi associated with biodeterioration in historical buildings from Barranquilla.

Samples	Phototrophic mic	roorganisms	Fungi	Aesthetic
	Phylum Cyanophyta	Phylum Chlorophyta	Division Ascomycota	alteration and biodeterioration
В1	Phormidium sp., Oscillatoria sp., Chroococcus sp., Cyanosarcina sp.,	-	Aspergillus niger, Aspergillus fumigatus, Penicillium sp., Fusarium sp.	Green, brown to black patinas and heavy discoloration, cracks, flakes off and swelling of walls. Black patinas, heavy discoloration and cracks on walls and balcony.
В2	Asterocapsa sp., Phormidium sp., Oscillatoria sp.	-	Aspergillus niger, Aspergillius fumigatus, Penicillium sp., Fusarium sp.	Black patinas and heavy discoloration on walls
В3	Chroococcus sp., Phormidium sp., Oscillatoria sp., Cyanosarcina sp., Gloeocapsa sp.	-	Aspergillus niger, Aspergillius fumigatus, Penicillium sp., Fusarium sp.	Black patinas, heavy discoloration and light flake off on walls.
В4	Chroococcus sp. Phormidium sp., Leptolyngbya sp., Lyngbya sp.	-	Aspergillus niger, Aspergillus fumigatus, Penicillium sp.	Green, brown to black patinas and heavy discoloration, cracks, flakes off and swelling of walls.
В5	Phormidium sp., Oscillatoria sp., Gloeocapsa sp., Lyngbya sp.	Chlorella sp., Trentepohlia sp.	Aspergillus niger, Aspergillus fumigatus, Penicillium sp., Fusarium sp.	Black patinas and heavy discoloration on walls

Stichococcus bacillaris in substrates of urban buildings and monuments (Gallego-Cartagena et al., 2020; Vojtková, 2017; Crispim et al., 2003; Gaylarde and Gaylarde, 2002).

3.2. Mineralogical characterization of the constructions materials

In the mortars analyzed by XRD, the major crystalline phases identified were quartz (α -SiO₂) and calcite (CaCO₃), although, in samples B1, B3, and B5, plagioclases (NaAlSi₃O₈-CaAl₂Si₂O₈) and potassium feldspars (KAlSi₃O₈) were also identified (Table S2). According to the XRD results, these materials can be related to calcareous mortars. Tremolite $(Ca_2Mg_5Si_8O_{22}(OH)_2)$, dolomite $(CaMg(CO_3)_2)$, and rutile $(\alpha$ -TiO₂) were also detected in sample B2, while gypsum (CaSO₄·2H₂O) in sample B4. The presence of tremolite (Ca₂Mg₅Si₈O₂₂(OH)₂) in building materials is associated with fibrous materials related to asbestos (for example, stucco with asbestos), still used as feedstock in the preparation of filler for decorative coating, protection, and sound-isolating on the façades of buildings in some countries (Paglietti et al., 2016; Bloise et al., 2008). It should be highlighted that during the sample preparation, when grinding B2 sample before XRD analysis, a mechanical behavior of plastic/ductile nature was observed in the sample. The presence of rutile $(\alpha\text{-TiO}_2)$ in B2 sample could be related to the paint layer applied on the surface of the facade. The identification of gypsum in sample B4 could also be related with its intentional addition in the mortar or due to a sulfation process of the calcium carbonate of the mortar.

3.3. Elemental composition of the construction materials

EDXRF analysis of the internal part (not exposed to the atmosphere) and the external part (exposed to the atmosphere) of the mortars allowed to detect major elements such as Mg, Al, Si, P, S, Cl, K, Ca and Fe and minor elements such as Ti, Mn, Zn, Rb, Sr, Pb (Table 2).

The high concentrations of Ca and Si in the samples are related to calcite and quartz presence resulting from the fine aggregate of the mortar. The highest concentrations of Al, Si, and K are related to the aggregates of mortars rich in aluminosilicates (plagioclase and feldspars).

Significant differences in the concentration of some major elements were observed comparing the external and internal parts (Tables 2 and 3). Remarkable is the higher concentration of Fe (24.0 \pm 3.0%) in the external part of the mortar of B5 sample, which shows an increase of concentration up to 72%. In this sample, a blackish dust was observable on the external part of the mortar, which could be associated to depositions of airborne particulate matter rich in Fe. In general terms, the increase of Fe comparing the external and internal parts of the mortars from all the samples represents a 62 \pm 7% as an average value. Moreover, external parts of the mortar in sample B2 and B4 are especially rich in Ti (concentrations of 24.6 \pm 3.1 and 32.7 \pm 4.1 % respectively) due to the use of TiO2 as coating in these external mortars/plasters.

Zn was under the Limit of Quantification in the internal part of all the mortar samples. However, in the external parts, its concentration range from 0.1 to 0.01% depending on the sample (Tables 2 and 3).

Pb was not detected or it is set below the Limit of Quantification in the internal parts of all the samples. On the contrary, in the external parts of samples B1 and B2, this heavy metal was detected at concentrations higher than 0.01%. The same trend was observable for Mn in the external parts of mortar samples B1 and B5 with increments of 76 and 100% respectively. The increase of chlorine in the external parts of samples B2 and B5 could be related to the diffuse influence of the marine aerosol or to an irregular distribution of Cl in the mortar due to capillary rise of water rich in chlorides (Table 2).

3.4. Study of the nature and distribution of the airborne particulate matter rich in metals in the SABs through elemental imaging

The elemental imaging study by μ -EDXRF allowed to determine the distribution of metals accumulated on the SAB. To start, direct imaging analyses were conducted, being able to detect the following elements in all the SABs: Mg, Al, Si, P, S, K, Ca and Fe as major elements; and Ti, Mn, Zn, Rb, Sr, Pb, Cu, V, Cr and Ni, as minor/trace elements up to some μ g/g (Figs. 3 and 4; Table 4). As a SAB can be considered a light material, due to the penetration ability of the X-Rays, to ensure that the mortar below the SAB do not contribute to the final result, SAB samples B1 and B4 (considering the three samples extracted from each location) were selected for an additional test. In these samples, the SABs were separated from the mortar scraping and separating their mass. The isolated mass of the SAB samples was independently analyzed through a multi-point strategy being possible to detect the same elements identified through the direct measurements of the SABs on the mortars (Table S3 and Fig. 4).

The semiquantitative results obtained in the scraped SABs indicate that P and S concentrations were lower in sample B4 (0.82 \pm 0.15%) compared with the sample B1 (8.54 \pm 3.21%) (Table S3). The same was observed for S (0.88 \pm 0.29% in B4 vs. 8.00 \pm 1.83% in B1) (Table S3). As P and S are representative elements of the constitution of biological matrices, the low concentration of these elements in B4 suggests that it cannot be considered a structurally constituted SAB although biocolonizers were identified on it. Indeed, at the naked eye, the crust does not show a filamentous structure (Fig. 2c). The especially high concentration of specific heavy metals (Ti, Cr, Fe, Ni, Zn, Pb) (Table S3) in B4 suggests that this crust has been formed due to the deposition of atmospheric airborn particulate matter (particles coming from soiling

Table 2
Concentration of major elements (norm. wt.%) in the external or exposed parts (EP) and internal or non-exposed parts (IP) of the mortar samples without the influence of the SABs.

Samples	Mg		Al		Si		P		S		Cl		K		Ca	a	Fe	
	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP
B1	<loq< td=""><td><loq< td=""><td>9.7</td><td>2.5</td><td>40.7</td><td>10.5</td><td><loq< td=""><td><loq< td=""><td>1.3</td><td>0.3</td><td><loq< td=""><td>N.D</td><td>2.3</td><td>0.7</td><td>40.7</td><td>85.2</td><td>4.6</td><td>0.7 ±</td></loq<></td></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td>9.7</td><td>2.5</td><td>40.7</td><td>10.5</td><td><loq< td=""><td><loq< td=""><td>1.3</td><td>0.3</td><td><loq< td=""><td>N.D</td><td>2.3</td><td>0.7</td><td>40.7</td><td>85.2</td><td>4.6</td><td>0.7 ±</td></loq<></td></loq<></td></loq<></td></loq<>	9.7	2.5	40.7	10.5	<loq< td=""><td><loq< td=""><td>1.3</td><td>0.3</td><td><loq< td=""><td>N.D</td><td>2.3</td><td>0.7</td><td>40.7</td><td>85.2</td><td>4.6</td><td>0.7 ±</td></loq<></td></loq<></td></loq<>	<loq< td=""><td>1.3</td><td>0.3</td><td><loq< td=""><td>N.D</td><td>2.3</td><td>0.7</td><td>40.7</td><td>85.2</td><td>4.6</td><td>0.7 ±</td></loq<></td></loq<>	1.3	0.3	<loq< td=""><td>N.D</td><td>2.3</td><td>0.7</td><td>40.7</td><td>85.2</td><td>4.6</td><td>0.7 ±</td></loq<>	N.D	2.3	0.7	40.7	85.2	4.6	0.7 ±
			$\pm \ 2.0$	\pm	\pm 5.1	± 1.9			\pm	\pm			\pm	\pm	\pm	\pm	$\pm~0.8$	0.2
				0.5					0.3	0.2			0.5	0.3	4.9	10.0		
B2	$5.2 \pm$	7.0 \pm	5.7	1.2	18.5	4.8	$0.9 \pm$	<loq< td=""><td>1.4</td><td>0.7</td><td>2.4 \pm</td><td><loq< td=""><td>0.9</td><td>0.1</td><td>37.9</td><td>86.2</td><td>2.4</td><td><loq< td=""></loq<></td></loq<></td></loq<>	1.4	0.7	2.4 \pm	<loq< td=""><td>0.9</td><td>0.1</td><td>37.9</td><td>86.2</td><td>2.4</td><td><loq< td=""></loq<></td></loq<>	0.9	0.1	37.9	86.2	2.4	<loq< td=""></loq<>
	0.9	1.6	± 0.9	\pm	± 1.9	$\pm~0.8$	0.4		\pm	\pm	0.2		\pm	\pm	\pm	\pm	$\pm~0.2$	
				0.8					0.3	0.2			0.1	0.0	1.8	10.0		
В3	<loq< td=""><td><loq< td=""><td>10.6</td><td>9.7</td><td>34.1</td><td>36.9</td><td><loq< td=""><td><loq< td=""><td>1.1</td><td>0.7</td><td><loq< td=""><td><loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<></td></loq<></td></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td>10.6</td><td>9.7</td><td>34.1</td><td>36.9</td><td><loq< td=""><td><loq< td=""><td>1.1</td><td>0.7</td><td><loq< td=""><td><loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<></td></loq<></td></loq<></td></loq<></td></loq<>	10.6	9.7	34.1	36.9	<loq< td=""><td><loq< td=""><td>1.1</td><td>0.7</td><td><loq< td=""><td><loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td>1.1</td><td>0.7</td><td><loq< td=""><td><loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<></td></loq<></td></loq<>	1.1	0.7	<loq< td=""><td><loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<></td></loq<>	<loq< td=""><td>3.0</td><td>2.3</td><td>45.4</td><td>47.0</td><td>4.5</td><td>$2.7~\pm$</td></loq<>	3.0	2.3	45.4	47.0	4.5	$2.7~\pm$
			± 3.0	\pm	\pm 8.0	\pm 5.8			\pm	\pm			\pm	\pm	\pm	\pm 4.0	$\pm~0.2$	0.7
				1.9					0.1	0.1			0.6	0.6	1.7			
B4	$2.5~\pm$	$3.5 \pm$	18.7	5.0	18.2	15.3	<loq< td=""><td>0.4 \pm</td><td>0.7</td><td>2.1</td><td>$0.5 \pm$</td><td>$1.0 \pm$</td><td>0.5</td><td>0.6</td><td>24.8</td><td>67.6</td><td>1.2</td><td>$3.7~\pm$</td></loq<>	0.4 \pm	0.7	2.1	$0.5 \pm$	$1.0 \pm$	0.5	0.6	24.8	67.6	1.2	$3.7~\pm$
	0.8	1.5	\pm 3.5	\pm	± 1.6	\pm 4.9		0.2	\pm	\pm	0.1	0.4	\pm	\pm	\pm	$\pm~9.0$	$\pm~0.2$	0.8
				1.4					0.1	0.9			0.1	0.1	3.1			
B5	$5.2 \pm$	<loq< td=""><td>8.7</td><td>3.9</td><td>16.3</td><td>5.8</td><td>$0.3 \pm$</td><td><loq< td=""><td>0.3</td><td>0.5</td><td>$0.5 \pm$</td><td>N.D</td><td>1.2</td><td>0.3</td><td>42.9</td><td>82.4</td><td>24.0</td><td>$6.8 \pm$</td></loq<></td></loq<>	8.7	3.9	16.3	5.8	$0.3 \pm$	<loq< td=""><td>0.3</td><td>0.5</td><td>$0.5 \pm$</td><td>N.D</td><td>1.2</td><td>0.3</td><td>42.9</td><td>82.4</td><td>24.0</td><td>$6.8 \pm$</td></loq<>	0.3	0.5	$0.5 \pm$	N.D	1.2	0.3	42.9	82.4	24.0	$6.8 \pm$
	1.3		± 1.3	\pm	± 1.5	$\pm~0.4$	0.1		\pm	\pm	0.1		\pm	\pm	\pm	±	\pm 3.0	1.4
				0.4					0.1	0.0			0.4	0.1	5.2	10.0		

<LOQ: Under the Limit of Quantification N.D: Non-detected.

Errors are expressed as one time the standard deviation of the mean concentration calculated from the values obtained from the three individual samples from each sampling location.

Table 3

Concentration of minor/trace elements (norm. wt.%) in the external or exposed parts (EP) and internal or non-exposed parts (IP) of the mortars samples without the influence of the SABs.

Samples	Ti			Mn		Zn		Rb		Sr		Pb	
EP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	EP	IP	
B1	0.6 ± 0.2	0.06 ± 0.02	0.14 ± 0.11	0.03 ± 0.01	0.01 ± 0.01	<loq< td=""><td>0.006 ± 0.002</td><td>N.D</td><td>0.12 ± 0.08</td><td>0.02 ± 0.01</td><td>0.02 ± 0.01</td><td>N.D</td></loq<>	0.006 ± 0.002	N.D	0.12 ± 0.08	0.02 ± 0.01	0.02 ± 0.01	N.D	
B2	$\begin{array}{c} \textbf{24.6} \; \pm \\ \textbf{3.1} \end{array}$	$\begin{array}{c} 0.04 \pm \\ 0.01 \end{array}$	$\begin{array}{c} \textbf{0.04} \pm \\ \textbf{0.01} \end{array}$	$\begin{array}{c} 0.06 \; \pm \\ 0.02 \end{array}$	0.08 ± 0.03	<loq< td=""><td><loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$</td><td>$\begin{array}{c} \textbf{0.03} \pm \\ \textbf{0.01} \end{array}$</td><td>$\begin{array}{c} \textbf{0.04} \pm \\ \textbf{0.01} \end{array}$</td><td>N.D</td></loq<></td></loq<>	<loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$</td><td>$\begin{array}{c} \textbf{0.03} \pm \\ \textbf{0.01} \end{array}$</td><td>$\begin{array}{c} \textbf{0.04} \pm \\ \textbf{0.01} \end{array}$</td><td>N.D</td></loq<>	N.D	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	$\begin{array}{c} \textbf{0.03} \pm \\ \textbf{0.01} \end{array}$	$\begin{array}{c} \textbf{0.04} \pm \\ \textbf{0.01} \end{array}$	N.D	
В3	1.0 ± 0.2	0.4 ± 0.1	$\begin{array}{c} 0.13 \pm \\ 0.07 \end{array}$	0.09 ± 0.03	0.02 ± 0.01	<loq< td=""><td><loq< td=""><td><loq< td=""><td>$\begin{array}{c} 0.11 \; \pm \\ 0.05 \end{array}$</td><td>$\begin{array}{c} \textbf{0.01} \pm\\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td><loq< td=""><td>$\begin{array}{c} 0.11 \; \pm \\ 0.05 \end{array}$</td><td>$\begin{array}{c} \textbf{0.01} \pm\\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td>$\begin{array}{c} 0.11 \; \pm \\ 0.05 \end{array}$</td><td>$\begin{array}{c} \textbf{0.01} \pm\\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<>	$\begin{array}{c} 0.11 \; \pm \\ 0.05 \end{array}$	$\begin{array}{c} \textbf{0.01} \pm\\ \textbf{0.01} \end{array}$	<loq< td=""><td><loq< td=""></loq<></td></loq<>	<loq< td=""></loq<>	
B4	$\begin{array}{c} \textbf{32.7} \pm \\ \textbf{4.1} \end{array}$	0.6 ± 0.1	<loq< td=""><td>$\begin{array}{c} 0.10\ \pm \\ 0.02 \end{array}$</td><td>$0.015 \pm 0.007$</td><td><loq< td=""><td><loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.30 \; \pm \\ 0.15 \end{array}$</td><td>$\begin{array}{c} \textbf{0.15} \pm \\ \textbf{0.02} \end{array}$</td><td><loq< td=""><td>N.D</td></loq<></td></loq<></td></loq<></td></loq<>	$\begin{array}{c} 0.10\ \pm \\ 0.02 \end{array}$	0.015 ± 0.007	<loq< td=""><td><loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.30 \; \pm \\ 0.15 \end{array}$</td><td>$\begin{array}{c} \textbf{0.15} \pm \\ \textbf{0.02} \end{array}$</td><td><loq< td=""><td>N.D</td></loq<></td></loq<></td></loq<>	<loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.30 \; \pm \\ 0.15 \end{array}$</td><td>$\begin{array}{c} \textbf{0.15} \pm \\ \textbf{0.02} \end{array}$</td><td><loq< td=""><td>N.D</td></loq<></td></loq<>	N.D	$\begin{array}{c} 0.30 \; \pm \\ 0.15 \end{array}$	$\begin{array}{c} \textbf{0.15} \pm \\ \textbf{0.02} \end{array}$	<loq< td=""><td>N.D</td></loq<>	N.D	
В5	0.9 ± 0.3	$\begin{array}{c} 0.27 \; \pm \\ 0.03 \end{array}$	$\begin{array}{c} \textbf{0.03} \pm \\ \textbf{0.02} \end{array}$	<loq< td=""><td>0.03 ± 0.01</td><td><loq< td=""><td><loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.06 \; \pm \\ 0.03 \end{array}$</td><td>$\begin{array}{c} \textbf{0.07} \pm \\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<></td></loq<></td></loq<>	0.03 ± 0.01	<loq< td=""><td><loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.06 \; \pm \\ 0.03 \end{array}$</td><td>$\begin{array}{c} \textbf{0.07} \pm \\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<></td></loq<>	<loq< td=""><td>N.D</td><td>$\begin{array}{c} 0.06 \; \pm \\ 0.03 \end{array}$</td><td>$\begin{array}{c} \textbf{0.07} \pm \\ \textbf{0.01} \end{array}$</td><td><loq< td=""><td><loq< td=""></loq<></td></loq<></td></loq<>	N.D	$\begin{array}{c} 0.06 \; \pm \\ 0.03 \end{array}$	$\begin{array}{c} \textbf{0.07} \pm \\ \textbf{0.01} \end{array}$	<loq< td=""><td><loq< td=""></loq<></td></loq<>	<loq< td=""></loq<>	

<LOQ: Under the Limit of Quantification N.D: Non-detected.

Errors are expressed as one time the standard deviation of the mean concentration calculated from the values obtained from the three individual samples from each sampling location.

and anthropogenic activities). Among all the samples, only in B4 lead was detected as part of the deposited crust.

In general, SAB samples showed the presence of heavy metals such as Cu, V, Cr, and Ni, not detected in the mortar itself (Tables 2–3). μ-EDXRF imaging revealed that S is homogeneously distributed (Fig. 3) in the SABs, confirming its relation with the structure of the biocolonizers. In contrast, Fe and Ti (specially in samples B2 and B5), Mn (in B1, B3 and B4), Cr (in B2 and B4), Zr (in B1 and B5) are heterogeneously distributed as hot spots on the mentioned SABs. Furthermore, a similar distribution was detected for elements such as Ni, in B4 and Zn in B2, B3 and B5 (Fig. 3). Moreover, B4 sample was the only one in which Pb was detected (Table S3 and Fig. 3) in very punctual areas, confirming that this heavy metal is part of the PM that forms the previously mentioned crust.

3.5. SEM-EDS study of single particles and clusters accumulated in the SABs

In order to examine the elemental composition of single-particles and clusters (aggregated particles) at a lower lateral resolution, single point and imaging SEM-EDS studies were also conducted. 128 particles in total were studied on the SABs from all construction (Table 5, Table S4, Figs. 5–7 and Fig. S9).

71% of the total particles analyzed belong to individual ones with diameters ranging from 2 to 40 $\mu m,$ while the particles that form clusters

correspond to 29% with diameters from 20 to 80 μ m (Table 5 and Table S4). 56% of the particles are in the range of PM₁₀ and 21% are close to PM_{2.5} size, a high percentage considering that the lower size particles are the ones that could generate adverse effects for human health.

The highest number of individual particles were identified in SABs B1 and B2. A higher percentage of clusters were detected in B1, B2 and B3. Heavy metals such as Fe, Mn, Ni, Cr and Ti were identified on individual particles, with sizes ranging from 5 to 40 μm (Table 5, Figs. 5–7 and Fig. S9d). This result agrees with the identification of hot spots of these metals by μ -EDXRF (Fig. 3). Specifically, Fe, Ti and Ni were detected in a higher proportion in B1 and B2 samples (Table 3). Ti-rich particles detected by SEM-EDS on the SABs could be related to the TiO₂ coating paint that has flaked off the building material or to activities related to iron welding and auto mechanic workshops that take place near the constructions where the SABs were extracted (Andrade-Guel et al., 2022). In addition, Fe-Ti-Mn-rich spherical particles were occasionally observed in the SAB filaments (Fig. 5b-d and Fig. S9). These particles are possibly emitted by the metallurgical industries and the combustion of petrochemical products derived from vehicular traffic (fuel oil, automobile oil) (Wang et al., 2021; Mitsos et al., 2022).

In the cluster particles, elements such as Al, Si, Ca, Mg, Na, Cl, Fe, K, Ti, Cu, Zn, Ba, Pb, Cr, Mn, Br, Ni, Sn, Sb and V were detected (Table 5). As bromine was not observable by μ -EDXRF, its presence can be

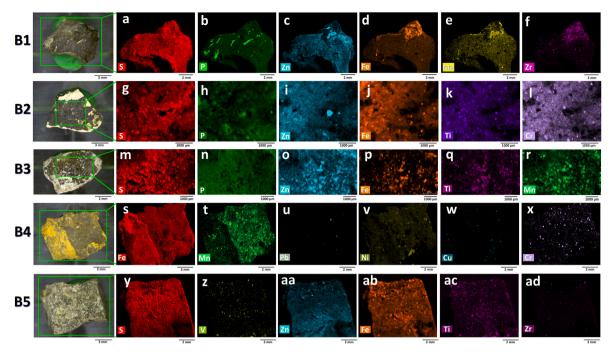


Fig. 3. μ-EDXRF imaging of SAB samples. Elemental distribution maps of sample B1: a) S b) P c) Zn d) Fe e) Mn f) Zr Elemental distribution maps of sample B2: g) S h) P i) Zn j) Fe k) Ti l) Cr. Elemental distribution maps of sample B3: m) S n) P o) Zn p) Fe q) Ti r) Mn. Elemental distribution maps of sample B4: s) Fe t) Mn u) Pb v) Ni w) Cu x) Cr. Elemental distribution maps of sample B5: y) S z) V aa) Zn ab) Fe ac) Ti ad) Zr.

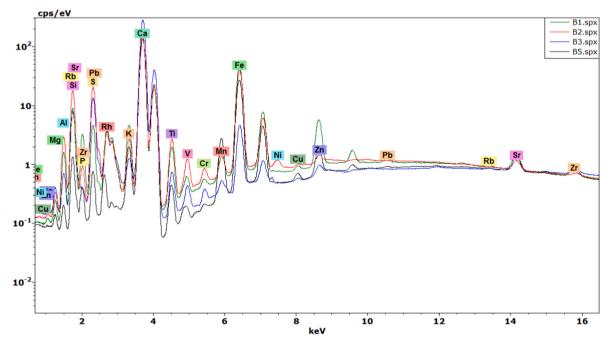


Fig. 4. Representative XRF spectra obtained from SAB samples.

considered punctual. In Figs. 5b, 6 and 7b, 7d and S9b examples of possible clustering of particles coming from geogenic or soiling of minerals/stones (Ca, Al and SO) and those emitted in anthropogenic activities (metallic particles rich in Ti, V, Cr, Ba, Fe and Ni) are observable. The diffuse influence of the marine aerosol was also confirmed by the identification of NaCl particles (see as an example in Fig. 5a, the particle deposited 8–10 μ m diameter in the filamentous structure of the SAB of sample B1). The punctual presence of this kind of particles and their size, which is set under the lateral resolution of the μ -EDXRF spectrometer explains the non-detection of chlorine in B1 by

this last mentioned technique.

In Fig. 6, an additional example of a cluster detected in the SAB of sample B2 is presented. Elements associated to crustal particulate matter (Al, Si, K and Ca mainly) and smaller metallic particles (even with sizes < than 10 $\mu m)$ of Fe and Ni were detected. These two metals were also identified as hot spots in the same sample by $\mu\text{-EDXRF}$. Regarding Ni, this last has also been identified in clusters together with Al, Ca, Si, Ti, Fe, Mn, Mg and Cr in samples B3 and B5 (Figs. 6 and 7a-b; Figs. S9b and S9e and Table 5).

The identification of clusters including Na-Cl (Fig. 5a), elements

Table 4 Heavy metals concentrations as a function of the area mapped in each SAB by μ -EDXRF (μ g/g/mm² units).

SAB sample											
	Fe	Ti	Mn	Zn	Rb	Sr	Pb	Cu	V	Cr	Ni
B1	0.3010 ± 0.0053	0.0280 ± 0.0010	0.0104 ± 0.0009	$\begin{array}{c} 0.0271 \; \pm \\ 0.0012 \end{array}$	0.0003 ± 0.0001	0.0059 ± 0.0018	0.0023 ± 0.0005	0.0008 ± 0.0002	0.0008 ± 0.0001	0.0012 ± 0.0003	ND
B2	$\begin{array}{c} 0.4071 \; \pm \\ 0.0105 \end{array}$	$\begin{array}{c} 0.0662 \pm \\ 0.0027 \end{array}$	$\begin{array}{c} 0.0111 \; \pm \\ 0.0008 \end{array}$	$\begin{array}{c} 0.0037 \; \pm \\ 0.0014 \end{array}$	a	$\begin{array}{c} 0.0046 \ \pm \\ 0.0017 \end{array}$	$\begin{array}{c} \textbf{0.0005} \pm \\ \textbf{0.0001} \end{array}$	$0.0009 \pm \\ 0.0001$	$\begin{array}{c} 0.0120 \pm \\ 0.0012 \end{array}$	$\begin{array}{c} 0.0040 \pm \\ 0.0012 \end{array}$	0.0029 ± 0.0004
В3	$\begin{array}{c} 0.0406 \; \pm \\ 0.0035 \end{array}$	$\begin{array}{c} 0.0151 \; \pm \\ 0.0004 \end{array}$	$\begin{array}{c} 0.0029 \pm \\ 0.0010 \end{array}$	$\begin{array}{c} 0.0024 \; \pm \\ 0.0009 \end{array}$	$\begin{array}{c} 0.0004 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0140 \; \pm \\ 0.0020 \end{array}$	$\begin{array}{c} 0.0002 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0005 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0034 \; \pm \\ 0.0006 \end{array}$	$\begin{array}{c} 0.0015 \pm \\ 0.0004 \end{array}$	ND
В5	$\begin{array}{c} 0.1801 \; \pm \\ 0.0023 \end{array}$	$\begin{array}{c} 0.0073 \; \pm \\ 0.0005 \end{array}$	$\begin{array}{c} 0.0207 \; \pm \\ 0.0021 \end{array}$	$\begin{array}{c} 0.0066 \; \pm \\ 0.0018 \end{array}$	$\begin{array}{c} 0.0002 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0055 \; \pm \\ 0.0018 \end{array}$	$\begin{array}{c} 0.0006 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0009 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0004 \; \pm \\ 0.0001 \end{array}$	$\begin{array}{c} 0.0003 \pm \\ 0.0001 \end{array}$	ND

Average mapped areas calculated from the three samples extracted from each sampling location (B1: $38.0 \pm 1.1 \text{ mm}^2$, B2: $17.8 \pm 0.8 \text{ mm}^2$, B3: $21.9 \pm 1.9 \text{ mm}^2$ and B5: $45.5 \pm 2.1 \text{ mm}^2$).

coming from crustal particles (Si, Ca, etc.) and other metals is a clear example of the combine diffuse influence of marine aeorosol, geological influence and anthropogenic activity (see cluster particles of Ti–Fe–Cr and Si–Fe–Cu in sample B1, Fig. 5b and Fig. S10a). Moreover, a combined influence of crustal particles and those emited from anthropogenic sources is clearly visible, as an example mention the detection of Ca–Ni in samples B2 and B3 (Fig. 7a and Fig. S9b), Si–Ca–Ti–Fe in sample B5 (Fig. 7e), Ca–Al–Si–K –Ni–Fe in sample B2 (Fig. 6), Fe–Mn–Cr in sample B4 (Fig. 7b), Si–Ca–Ni–Fe in sample B5 (Fig. S9e), S–Ba–Ca in sample B5 (Fig. 7d), and spherical particles of Fe with Ca contribution in sample B3 (Fig. 5d). All the metals detected corresponde to the ones identified by imaging EDXRF.

It is also remarkable that S and Ba were detected in all samples, although with a higher order of magnitude (<1%) in sample B2, B3 and B5 (Table 5). Particles rich in S–Ba could be related to barium sulfate, which is produced by the wear of automobile brakes, including the abrasion of the material of the brake pads and brake discs, caused by the grinding of the components of the brakes pads or the volatilization and condensation of the brake pad material (Carrero et al., 2013, 2014; Zafra et al., 2016; Pant and Harrison, 2013).

3.6. Correlation of the main anthropogenic emission sources according to the main metals accumulated in the SABs extracted from each construction

To extract additional information about the main metals accumulated on the SABs, and the possible emission sources, as a function of the location of the construction in the city of Barranquilla, the concentration of metals as a function of the mapped area of the SAB by μ –EDXRF was calculated (Table 4). As sample B4 was not defined as a SAB, it was not considered in this case. Moreover, results associated to heavy metals detected are only presented, the values associated to lighter elements have been removed from Table 4 for a better comprehension.

In this sense, the SAB B1 shows the highest concentration of Pb $(0.0023\pm0.0005~\mu g/g/mm^2)$ and high concentration of Fe $(0.3011\pm0.0053~\mu g/g/mm^2)$ and Cr $(0.0012\pm0.0003~\mu g/g/mm^2)$. The first two heavy metals can be associated to the emissions coming from road traffic. Some researches indicate that fine inhalable particles with diameters lower than 1 μm , rich in Fe and Pb, are generated in high temperatures processes derived from vehicular transport; among these, the burning of fossil fuels, tire and engine wear (Machado et al., 2008; Liu et al., 2014). The construction from which B1 sample was obtained is located in a locality of the metropolitan area highly influenced by a continuous traffic, specially coming from public transport. Moreover, this construction presents a poor conservation state due to lack of maintenance and restoration actions.

The SAB B2 corresponds to a construction that presents a relatively better state of conservation. In this case, the highest concentration of Fe (0.4071 \pm 0.0105 $\mu g/g/mm^2$) and Ti (0.0662 \pm 0.0027 $\mu g/g/mm^2$) were registered. Moreover, this was the only SAB that showed the

presence of Ni (0.0029 \pm 0.0004 $\mu g/g/mm^2$) (Table 4). The high concentrations of such metals determined by μ -EDXRF in this sample agree with the elemental composition of the single particles (2–12 μm) and cluster (7-80 µm) studied by SEM-EDS in this sample. Additionally, apart from Fe, Zn is also widely present in this sample. These last elements, together with Ni can be associated with fine particle emissions from continuous traffic generated by public transport (Fig. 3, Tables 4 and 5). Additional authors (Osorio-Martinez et al., 2021; Vianna et al., 2011; Poulakis et al., 2015) have proposed that the fine and coarse fractions of PM rich in metals such as Ti, Mn, Cu, V and Ni come from emissions from mixed anthropogenic activities (industry and transport) or geogenic (soil and urban dust) on tropical citys. All this elements are present in the SAB B2. It should be considered that the construction is located 800-1000 m away from emission sources related to the city's food and metallurgical industry and the vehicular flow related to mass public transportation routes.

Sample B3 is especially rich in Fe $(0.0406 \pm 0.0035 \ \mu g/g/mm^2)$, V $(0.0034 \pm 0.0006 \ \mu g/g/mm^2)$ and Cr $0.0015 \pm 0.0004 \ \mu g/g/mm^2)$ (Table 4). EDXRF study of the Sr concentration in the external vs internal part of mortars revealed an increment of this element in the external part (90%) versus the internal part (10%) (Table 3). The construction from which SAB B3 sample was extracted host the annual celebration of the carnivals of Barranquilla, in which fireworks shows take place. Unusual levels of Sr and Rb have been reported in fireworks aerosols (these metals together a Ba, Cu, Cr, Ti are used to provide silver and shiny effects) by different authors (Moreno et al., 2007; Hoyos et al., 2020; Pirker et al., 2021; Chatterjee et al., 2013; Schleicher et al., 2012). Therefore, the higher Sr presence in SAB B3 can be associated with this source.

Finally, sample B5 shows the lowest concentrations of metals per mm² for all the detected elements. This last agrees with the earliest contruction of the building in comparison with the rest buildings under study. Thefore, in this case, the SABs formed have been exposed less time to the environment, being able to accumulate a lower amount of particulate matter. In spite of that, the concentrations per area of Mn $(0.0207 \pm 0.0021~\mu g/g/mm^2~units)$, Fe $(0.1801 \pm 0.0023~\mu g/g/mm^2~units)$ and Ti $(0.0070 \pm 0.0005~\mu g/g/mm^2~units)$ (Table 4; Figs. 3 and 4; Fig. 7e) are quite high. This last can suggest that the façade of the construction from which this sample has been obtained could be mainly impacted by PM derived from activities related to metal manufacturing processes (for example, galvanizing, crushing, alloy) which are located in that sector of the city. Concretely, Mn could be related to the production of Hadfield steel, which contains Mn at an average concentration of 14–15% (Gasik, 2004).

4. Conclusions

The results presented in this work demonstrate that, in the future, the use of SABs growing of constructions can be explored as a promising

a Lower than 0.0001.

Table 5Elemental composition of particles studied on SAB samples obtained by EDS as a function of their size.

SAB	Particle type	2	Elemental composition						
	Single particles (µm)	Cluster particles (µm)	Major (>1%)	Minor (1%– 0.1%)	Trace (<0.1%)				
B1	5		O, C, Fe, Si	Al, Ca, Cu	Zn, Sn, Pb, Ba, Cr				
	10		C, O, Ca, P, Si	Al, Fe, Mg, Rb, Sr, Ni	Zn,				
	4		O, C, Ca, Si	Mg	Mn				
	5		O, C, Si, Al, K	Cu	V, Pb				
		10	O, C, Si, Mg, Al, Fe, Ca	Na, K, Ba	Cr, Sn, Cl, V				
		10	O, C, Ca, P, Si	Al, Na, Cu, Fe	Cl, Pb				
	2		O, C, Ti, Fe, Si, Ca,	Al	Zn				
		20	O, Ti, Fe, Ca, Si, Al	Mg, Mn, K, Ba	Cr, Sb, Pb, V				
	4		O, C, N, Ti	Al, Si, Ca, P	Ni, Cu, Cl, K, Pb				
		18	O, C, Ca, Si, Al	Fe, Mg, Sr, K, Ti	Zn, Cu, Cr, Mn, Ni, Na, Sn				
	15		O, C, Ca, P	Si, Mg, Fe, Na, Al, Br,	Mn, Ti, K, Cr, Ni				
	8		С, О	Sr, Zr, Rb, Ca, Si, Al, Mg,	Fe, Rb, Cl				
	2		O, Ca, C, Nb, Si	Cu, Ti					
	3		O, C, Ca, Si, Al	Fe, Mg, Sr, K, Ti	P, Sr, Cu, Cr, Mn, Ni, Sn, Na				
	2		C, O, Ca, Fe, S, Si	Al	Ni, V				
	18		C, Cl, Na, O	Ca, Zn	Pb				
	2		O, C, Ti, Fe, Si, Ca, Al	Mn, Zn, V	Ва				
		5	O, C, Ca, Si, Al	Fe, Mg, K, Na, Cr, Ba	Mn, Cl, Pb				
B2		12	Ni, C, Ca, O, Si, Al	S, Br, Fe, Mg, Na	Mn, Ti, Ba, Cr, Pb				
	2		C, O, Si, Ca, Al, S, Fe	Sb	Cu, Cr, Ti				
	12		O, Si, Ca, Al, K, Mg	Ti, Na, Fe, Sn, Ni, Sb	Mn, V				
	4		O, C, Ca, Ti, Si, Al	Fe, Mg, K, S, Sb	Sn, Ni, Cu				
		9	O, Si, Ca, Al, Mg	K, Ti, Fe	Cl				
	3		Fe, C, Si	Ca, Cr, Al, Mn, S, Cu, Ni	P, Pb				
	6		Fe, C, O, Si	Al, Cr	Na, S, Ni, Ba, Ti				
	4		C, O, Ca, Si	Al, K	Cl, Ti				
		7	Fe, C, O, Si	Al, Ca, Cr, Mn, Ni	Sb Sn, Rb, Sr				
	8		Ba, C, S, O, Ca	Al, Fe, Cr, Mn	Br				
	2		C, O, Fe, Ca, Mg, Si	Al, Ni	Mn, V, Zn				
	8		C, O, Si, Ca, Mg, Al, K	Br, S	Ni, Ba, Pb, V				
	3	80	C, Fe, O, Si C, O, Ca, Si	Cr, Cu Al, S, Fe, Ni, K	Ba, Pb Mg, Na, Ti, Cl				

Table 5 (continued)

SAB	Particle type	2	Elemental co	Elemental composition					
	Single particles (µm)	Cluster particles (µm)	Major (>1%)	Minor (1%– 0.1%)	Trace (<0.1%)				
В3		40	Ba, C, O, S, Ca, Si	Fe	Pb				
	4		C, Fe, O, Ca	Si, Mg, S, Sb	Sn, Pb, Ti, Cr				
	10		O, C, Mg, Ca, Si, Fe	Al, Cr	V, Sb, Cu				
	10		O, Fe, C	Ca, Sb	Mn, Sn, Pb Ba				
		5	C, O, Ca, Ni, Br, K	Si, Al, Mg, Sn, Pb	Zn				
	6		O, C, S, Ba, Ca	Si, Cr	Pb, Mg, Zn, Rb				
		10	Ni, C, O, Ca	Si, Al, Fe	Sb, Pb, Zn, Cu				
		12	O, C, Ba, S, Ca	Si, Al, Pb	Zn, Sb				
B4		10	Fe, O, C, Cr, Si, Al, Ni	Mn, Ca, Ti, Cu, Rb	Pb, Zn				
	2		O, C, Fe, Si, Al, Mg, Ca	Ti, Ba, Mn	V				
		3	Fe, Cr, C, Ni, O, Si, Mn	Mg, Ca, Ti	V, Ba				
		5	Fe, O, C, Si, Al	Mg, Mn, Rb, Ba	Sr, Pb				
		7	O, Si, Al, Fe, Mg	Ca, K, Na, Cr	Sn, Ti, V				
	5		Fe, O, Si, Mn, Al, Ca	Na, Ti, Sr, Sb, Ba	Pb				
	8		O, Fe, Mn, C, Si	Al, Na, Ca, Ba	Sb				
	10		O, Fe, C, Si	Al, Ca, Ti, Ni, Pb	Sr, Rb				
В5	2		O, C, Ba, S	Mg, Si, Mn, Fe	Pb, Cr, Zr				
	2		O, C, Si, Ca, Ni	Al, Fe, K, Sb	Ti, Cr				
	3		O, Ca, Br, K, Si	Sn, Sb	Ti, Cr, Pb				
	4		O, C, Ca, Si, Ni	Al, Mg, Fe	V				
	25		O, Fe, C, Ca, Si	Mg, Al, Zn	V, Cu, Ti, Cr, Ba				
	2		O, C, Ca, Si, Ni, Al	Fe	Ti, V, Pb				
		3	O, Ca, Si, Al, S,	Fe, Mg, Cl, Ni, Zn	Pb				

alternative to biomonitor atmospheric airborne particulate matter rich in heavy metals. In the future, it would be necessary to carry out additional controlled experiments exposing SABs grown on mortars mockups to airborne particulate matter rich in heavy metals. This experiment will allow to quantitatively evaluate the capacity of the SABs to retain the mentioned particles. Apart from the study of the SABs, this work also demonstrates that a selective study of the external and internal (not exposed to the atmosphere) parts of a mortar can offer information about possible accumulations of atmospheric particulate matter in construction materials.

The analytical methology here presented based on the direct use of $\mu\text{-EDXRF}$ and SEM-EDS single point and imaging strategy has been successfully applied to extract a general overview of the size, nature and distribution of individual and cluster particles deposited on the SABs growing on constructions focusing the attention on the metallic nature of the deposited airborne particulate matter.

Elemental imaging studies of the SABs allowed not only to map the metallic accumulations on them but also to discriminate crusts rich in

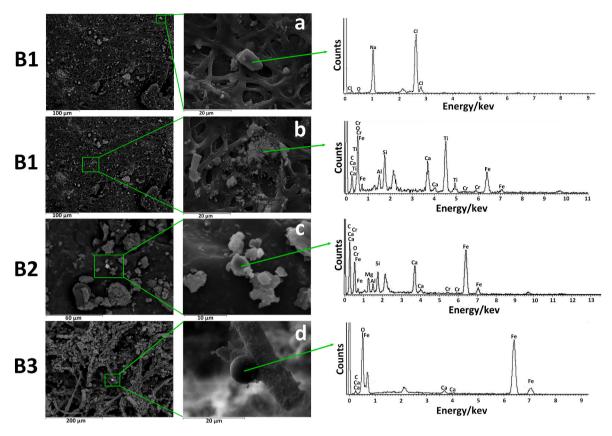


Fig. 5. SEM micrographs of airborne particulate matter deposited on SAB samples B1, B2 and B3 and EDS spectra acquired on the following selected particles: a) 10 μ m halite particle deposited on filamentous structures of the biocolonizers of B1. b) 12–18 μ m aggregates of titanium oxide and Ca/Fe/Al/Si/Cr deposited on filamentous structures of the biocolonizers of B1. c) 5–7 μ m aggregates of aluminosilicate and Fe/Cr deposited on the surface of B2. d) 8 μ m aggregates of iron oxide attached to biocolonizer filamentous structure of B3. Note: The non assigned signal in the spectra belongs to Au used to coat the samples.

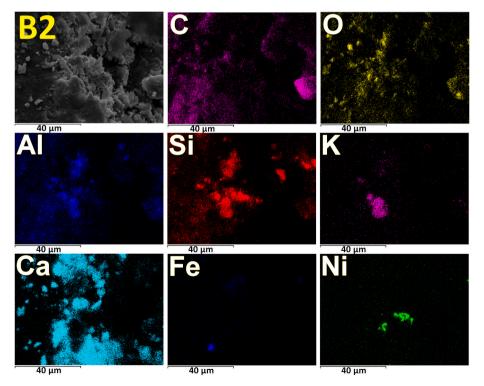


Fig. 6. Elemental distribution maps acquired by SEM-EDS of PM deposited on the SAB B2.

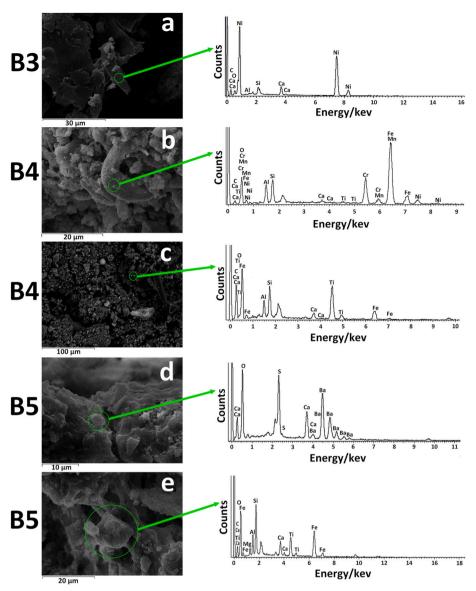


Fig. 7. SEM micrographs of airborne particulate matter deposited on SAB samples B3, B4 and B5 and EDS spectra acquired on the following selected particles. a) 10 μm aggregate of Ni and aluminosilicate deposited on the surface of B3. b) 10–12 μm aggregate iron manganese, aluminosilicate and Ti/Cr/Ni deposited on the surface of B4. c) 5–9 μm μm aggregate of aluminosilicate and Ti/deposited on the surface of B4. d) 3 μm aggregate of barium sulfate and Zn deposited on the surface of B5. e) 10–14 μm aggregates of titanium and iron oxides, aluminosilicate deposited on the surface of B5. Note: The non assigned signal in the spectra belongs to Au used to coat the samples.

heavy metals from real SABs. The semi-quantitative information per unit of area obtained from μ -EDXRF imaging studies of the SABs allowed also to correlate the main metals detected, with nearby anthropogenic impact sources (road traffic, impact of fireworks, impact of metallurgical industry and so on).

SEM-EDS was crucial to perform an in depth study at lower lateral resolution of individual particles or clusters deposited on the SABs. Apart from individual particles of higher dimensions (up to 40 μ m), clusters were also identified trapped on the SABs ranging sizes from 20 to 80 μ m. In agreement with the μ -EDXRF studies, the predominance of particulate matter rich in Zn, Fe, Mn, Ni and Ti was corroborated by SEM-EDS. This last technique allowed to identify trace metals (down to μ g/g), non-detectable by μ -EDXRF imaging such as Ba, Sb, Sn, Cl and Br (\leq 0,1%).

The obtained results demonstrate the usefulness of this methodology as a relevant tool for environmental monitoring, being a less tedious, time-consuming and environmentally sustainable alternative appropriate to extract quick conclusions helpful to assist stakeholders in environmental health and urban planning. Although this study has been focused in samples of reduced size, considering the availability of transportable macro-XRF scanners able to map bigger surfaces, this methodology can be transferred to bigger façades affected by SABs in

order to extract more representative results. This last would help to unequivocally classify SABs spontaneously growing on the façades of constructions as good alternatives to retain particulate matter rich in heavy metals acting as a roboust passive sampling alternative.

Author contributions statement

Euler Gallego Cartagena: Conceptualization, Data curation, Formal analysis, Investigation, Methodology, Software, Validation, Visualization, Roles/Writing - original draft, Writing - review & editing, Héctor Morillas: Investigation, Formal analysis, Software, Writing-review & editing. Wendy Morgado-Gamero: Writing - review & editing. Fabio Fuentes-Gandara: Roles/Writing - original draft. Víctor Vacca-Jimeno: Software, Validation. Isabel Salcedo: Visualization. Juan Manuel Madariaga: review & editing. Maite Maguregui: Conceptualization, Data curation, Methodology, Validation, Investigation, Formal analysis, Software, Writing-review & editing.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence

the work reported in this paper.

Data availability

The authors do not have permission to share data.

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Appendix A. Supplementary data

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