

# Metamaterials for Concentrated Solar Power applications

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- Human beings consume energy on their everyday life
- Traditionally, it has been produced using non-renewable resources such as fossil fuels or nuclear energy



Source: www.kidzworld.com



- This tendency is changing due to the development of renewable energies such as:
  - Wind Power





- This tendency is changing due to the development of renewable energies such as:
  - Hydropower



Source: www.emaze.com



- This tendency is changing due to the development of renewable energies such as:
  - Solar Power



Source: www.inhabitat.com



# Solar Energy

- However, there are many technologies inside the solar power:
  - Solar Thermal: Water is heated to be used as Domestic Hot Water (DHW), it's goal is to heat rooms through the heating, or be used as shower water no electricity is generated.



Source: www.mvmsolar.com



## Solar Energy

- However, there are many technologies inside the solar power:
  - Photovoltaics: DC electricity is generated from the use of semiconductors in these panels.



Source: www.iene.eu



# Solar Energy

- However, there are many technologies inside the solar power:
  - Concentrated Solar Power (CSP): Sun light is concentrated in a single point (or row) so that thermal energy is efficiently converted to electricity



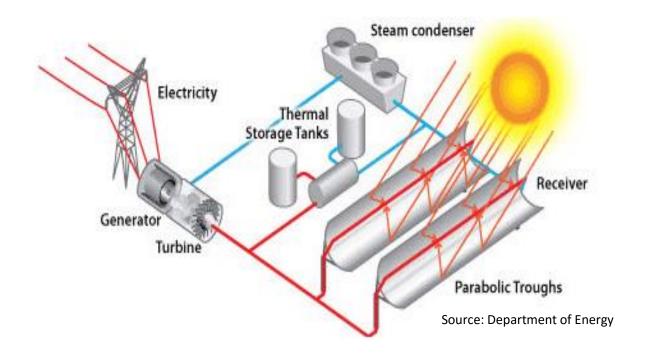
Source: www.pfenninger.com

Linear concentrator systems: parabolic



Source: Solar Energy Industries Association

- Linear concentrator systems: parabolic
  - Light is concentrated in a long pipe using a parabolic trough, where a liquid is heated and vaporized in order to the steam to move a turbine to produce electricity



Linear concentrator systems: linear Fresnel



Source: AREVA Solar

- Linear concentrator systems: linear Fresnel
  - Light is concentrated in a long pipe by using a number of long and thin movable flat mirrors, where a liquid is vaporized and moved towards a turbine to produce electricity.



Source: Open Source Ecology

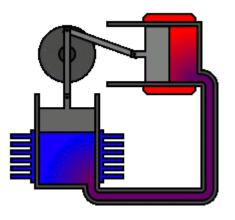
Dish/Engine systems: Stirling



Source: Wikimedia

- Dish/Engine systems: Stirling
  - Light is concentrated in a single point and air or another gas is heated. As it expands
    it moves a piston, the fluid is displaced and cooled down so the piston goes back to
    the initial position, generating the rotatory movement needed to generate
    electricity.



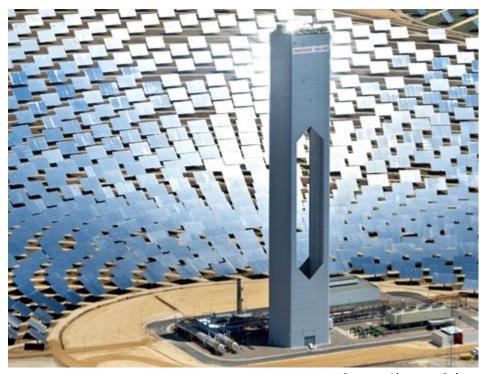


Credits: By Zephyris at the English language Wikipedia, CC BY-SA 3.0



# Types of CSP: High temperatures (T>400°C)

Solar Power Tower (SPT)

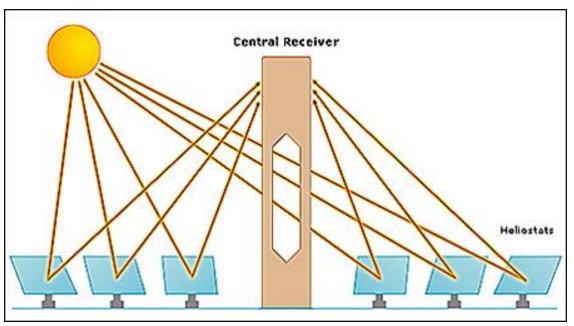


Source: Abengoa Solar



# Types of CSP: High temperatures (T>400°C)

- Solar Power Tower (SPT)
  - A number of mirrors (called heliostats) concentrate the light on top of a tall tower where water or salts are heated to create vapor water. This is later directed to the generator, where electricity is generated.



Source: SolarCellCentral.com



## Solar Power Tower

- The Solar Power Tower type of Concentrated Solar Power is still in its first development stages; the first ever created commercially oriented SPT was built in 2006 by Abengoa Solar in Spain.
- Several plants have been constructed since then, each one adding new improvements from what has been learnt from previous constructions.
- As a consequence, new SPT plants have been improving their production capacity, as can be seen in the yearly production of each of the currently active SPT's, on the following slide.



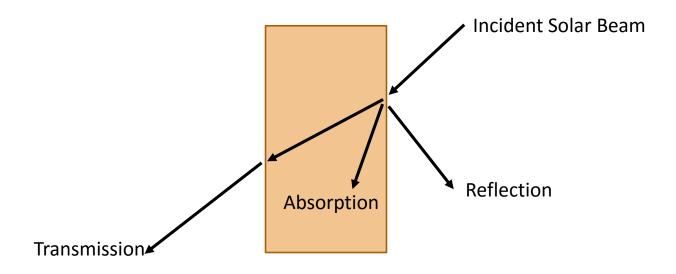
### Solar Power Tower

- To further understand what is actually happening on top of the SPTs, some basic notions of radiation are explained below.
- This notions will help understand where and how improvements can be done in currently active SPTs.



- When an incident light beam impacts a surface, three possible things can happen: the beam can be reflected, absorbed or transmitted
- Following Kirchoff's law, the sum of the three components must be the same as the incident beam:

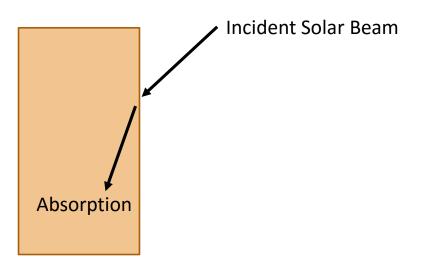
Incident Solar Beam = Absorption + Reflection + Transmission





#### Absorption:

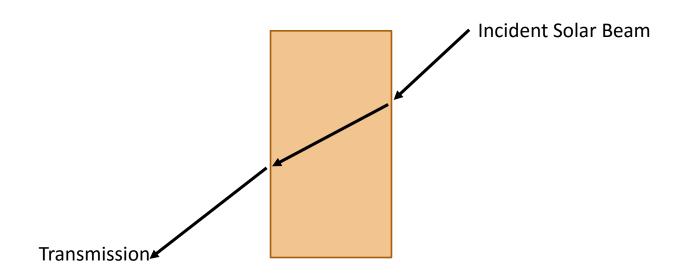
• Electrons on surfaces are vibrating at their natural frequency. If they are hit with a wavelength that vibrates in the same frequency, electrons absorb it and through movement transform it in thermal energy.





#### **Transmission:**

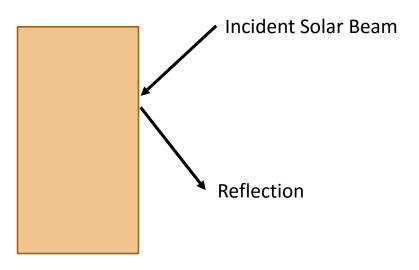
• When hitting wavelength does not match the natural frequency of the electrons, they vibrate for short periods of time. Then the energy is emitted as a wavelength. If the object is transparent, the process continues through the width of it, until it exits the object.





#### Reflection:

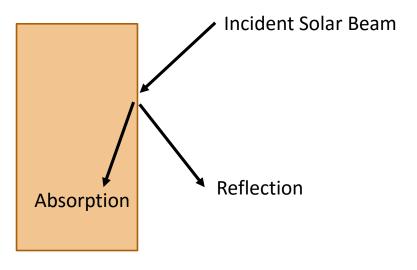
• When hitting wavelength does not match the natural frequency of the electrons and the object is opaque, the energy emitted as a wavelength is not absorbed by the atoms through the material.





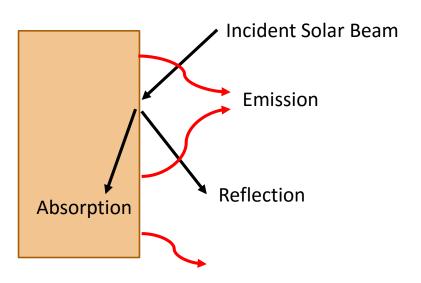
• In opaque objects, the total transmitted energy is zero, so Kirchoff's equation is as follows:

Absorption(%) + Reflection(%) = Incident Solar Beam(100%)





- In opaque objects, if the object has an absorption, its energy increases, thus its temperature raises
- Assuming the object is in steady state and therefore at constant temperature, it must emit everything that absorbs.

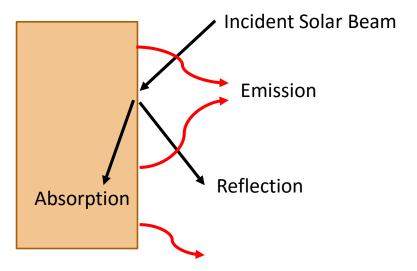




• Therefore, Emission = Absorption, turning Kirchoff's equation into

Emission(%) + Reflection (%) = Incident Solar Beam (100%)

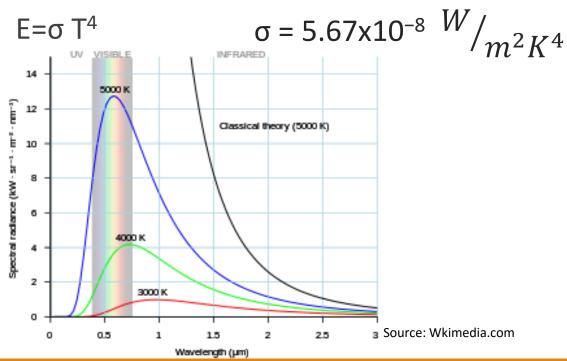
• Thus, **less** reflection = **more** emission





#### Blackbody:

 A blackbody is defined as a body can emit (and absorb!) the maximum theoretically possible amount of radiation, defined by Stefan Boltzmann's equation:





#### Emissivity:

- It is defined as the effectiveness of a material in emitting energy as thermal radiation. It is a factor that varies from 0 to 1, depending on how similar to a blackbody an object is.
- When placed inside the Stefan Boltzmann's equation, it shows the total radiation emitted by a body

$$E = \varepsilon \sigma T^4$$

- Examples:
  - Emissivity=1 → Same as Blackbody
  - Emissivity=0.4 → Emits 40% of a blackbody at that temperature
  - Emissivity=0 → Not emission at all



#### Greybody:

- It is defined as a body that has a constant value of emissivity through all wavelengths
- In case it varies due to roughness (reflection may be affected), it varies in the same way in all wavelengths



#### Non-greybody:

- This types of bodies are the ones that have a variable value of emissivity for different wavelengths
- Plastic bags for example are opaque to visible wavelengths, but transparent to infrared wavelengths





Source: physics.stackexchange.com



#### Thermal radiation:

 When a body is heated up, the particles inside it start moving. This thermal motion of charged particles generates an electromagnetic radiation that is emitted and can be measured.





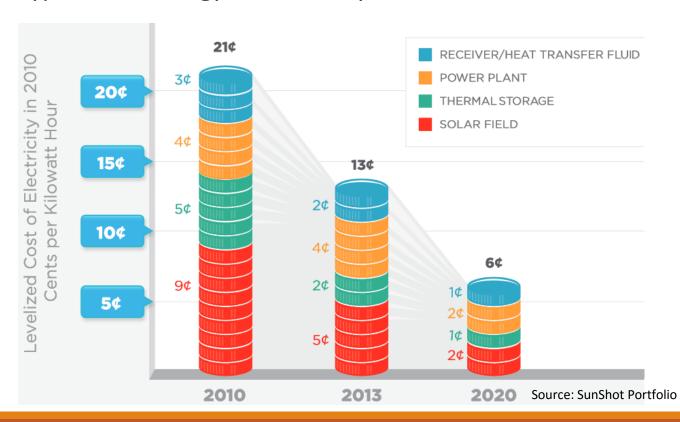
## Current coatings

- Nowadays, the most common coating being used on the receivers of the SPTs is the Pyromark® 2500 series paint, that have a solar absorptivity ( $\alpha$ ) of 0.96 and an emissivity ( $\epsilon$ ) of 0.86
- The absorptivity value is a really good one, but the emissivity still can be much lower to avoid large IR emissions.



## Current objectives

 The objective of the CSP industry is to lower the cost of the KWh of this type of technology to 6 cents by 2020





# Currently active Solar Power Tower plants

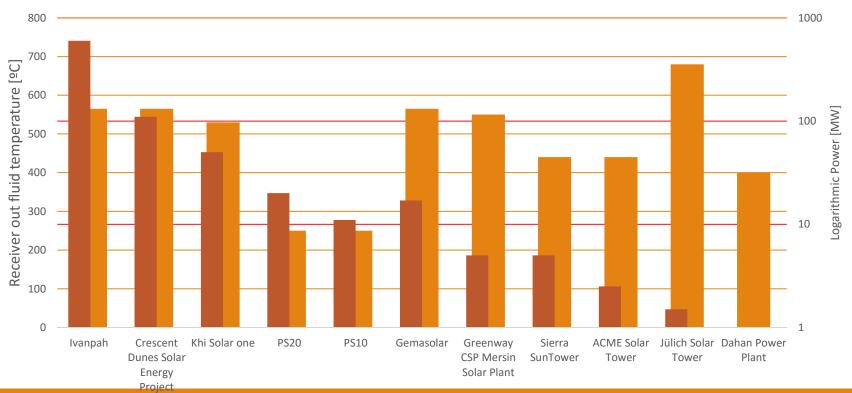
| Power plants                           | Installed capacity<br>(MW) | Yearly production (GWh) | Country      | Developer/<br>Owner                                   | Completed |
|--|----------------------------|-------------------------|--------------|---|-----------|
| Ivanpah Solar Power<br>Facility        | 600                        | 420                     | USA          | BrightSource Energy                                   | 2013      |
| Crescent Dunes Solar<br>Energy Project | 110                        | 500                     | USA          | SolarReserve  | 2013      |
| Khi Solar One                          | 50                         | -                       | South Africa | Abengoa   | 2016      |
| PS20 solar power tower                 | 20                         | 44                      | Spain        | Abengoa   | 2009      |
| Gemasolar                              | 17                         | 100                     | Spain        | Sener   | 2011      |
| PS10 solar power tower                 | 11                         | 24                      | Spain        | Abengoa   | 2006      |
| Jülich Solar Tower                     | 1.5                        |                         | Germany      |   | 2008      |
| Greenway CSP Mersin<br>Solar Plant     | 5                          |                         | Turkey       | Greenway CSP  | 2013      |
| Sierra SunTower                        | 5                          |                         | USA          | eSolar  | 2009      |
| ACME Solar Tower                       | 2.5                        |                         | India        | ACME Group  | 2011      |
| Dahan Power Plant                      | 1                          |                         | China        | Inst. of Elec. Eng. of Chinese<br>Academy of Sciences | 2012      |



## Current objectives

• To do so, the current objective for the hot fluid's exit temperature to be higher than 650°C.







## Current objectives

- On the other hand, the current objective for the thermal efficieny of the receiver is to be higher than 90%
- To achieve that a lot of research is being done in improving several aspects of the production
- An analysis of different efficiencies and their effects is shown in the following slides



# Efficiency

#### Four main efficiency values existing:

- Mirror efficiency: How efficiently light beams are reflected in the Heliostats
- Receiver efficiency: How efficiently solar irradiation is transformed into thermal energy
- Carnot efficiency: How efficiently thermal energy is turned into mechanical energy
- AC conversion efficiency: How efficiently mechanical energy is transformed into electrical energy



Relevance of the efficiencies to this project:

- Mirror efficiency: It is going to be ignored as it usually is around 90-97% and is not the objective of this research
- Receiver efficiency: Important and relevant to this research
- Carnot efficiency: Important and relevant to this research
- AC conversion efficiency: It is going to be ignored as it is not the objective of this research



#### Carnot efficiency:

- Measures the efficiency of thermal to mechanical transformation of energy
- The hotter the hot source and the colder the cold source the better efficiency
- Hot source's temperature is the temperature at which water or salts are heated on top of the tower, and the cold temperature is the temperature of the water or salts entering the tower to be heated

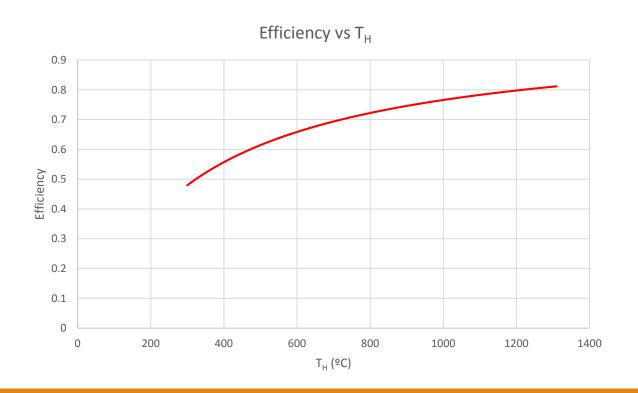
$$\eta_{th} \le 1 - \frac{T_C}{T_H}$$



#### Carnot efficiency:

Assuming T<sub>C</sub>=25°C

$$\eta_{th} \leq 1 - \frac{T_C}{T_H}$$





- It might seem logical then to put as many heliostats as possible to heat the fluid as much as possible
- However, as mentioned before, Carnot efficiency is not the only one playing an important role in the overall efficiency
- Receiver's efficiency complicates the calculation of the optimum working temperature, as the hotter a body is the more infrared radiation it emits
- The more IR emission a body has the faster it cools down!



#### Receiver's efficiency:

• It is defined as a function between the total solar radiation impacting the receiver on top of the tower ( $Q_{solar}$ ), the ammount of radiation actually absorbed by the receiver ( $Q_{absorbed}$ ) and the ammout of radiation the receiver emmits due to IR radiation ( $Q_{lost}$ )

• 
$$Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$$

• 
$$Q_{absorbed} = \alpha \cdot Q_{solar}$$

• 
$$Q_{lost} = A \cdot \varepsilon \cdot \sigma \cdot T_H^4$$

$$\gamma = \frac{Q_{absorbed} - Q_{lost}}{Q_{solar}}$$



Receiver's efficiency: Q<sub>solar</sub>

• Q<sub>solar</sub> is the total ammount of radiation that impacts the receiver, and it is defined as follows:

$$Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$$

• Where  $\eta_{optics}$  is the Mirror's efficiency ( $\approx$ 1), C is the Concentration value, A stands for the Receiver's area (in m²) and I is the Solar Irradiance (in W/m²)



Receiver's efficiency:  $Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$ 

• Concentration is a measure of how concentrated the sunlight is. The more Heliostats, the more concentrated the sunlight will be. It can be calculated as the ratio of the incoming radiation's area  $A_{\text{Heliostats}}$  to the outgoing radiation's area  $A_{\text{receiver}}^*$ 

$$C = \frac{A_{Heliostats}}{A_{Receiver}}$$

• Where  $A_{Heliostats}$  is the total area of the installed heliostats and  $A_{Receiver}$  is the area of the receiver on top of the tower

<sup>\*</sup>Robert Pitz-Paal, High Temperature Solar Concentrators, Solar Energy Conversion And Photoenergy Systems, Vol I



Receiver's efficiency:  $Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$ 

Using Plack's law to calculate the Spectral Irradiance (I),

$$B_{\lambda}(\lambda, T) = \frac{2hc^2}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda k_B T}} - 1}$$

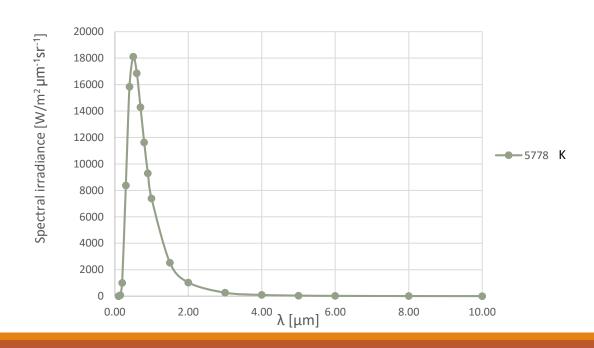
- Where h is the Plack constant,  $K_{\text{B}}$  the Boltzmann constant, c the speed of light,  $\lambda$  is the wavelength and T temperature
- Spectral Irradiance's SI units are W m<sup>-3</sup> sr<sup>-1</sup>



Receiver's efficiency:  $Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$ 

Assuming Sun's surface is at 5778K,

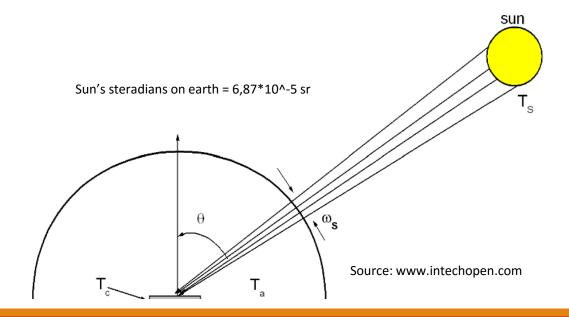
Sun's blackbody radiation





Receiver's efficiency:  $Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$ 

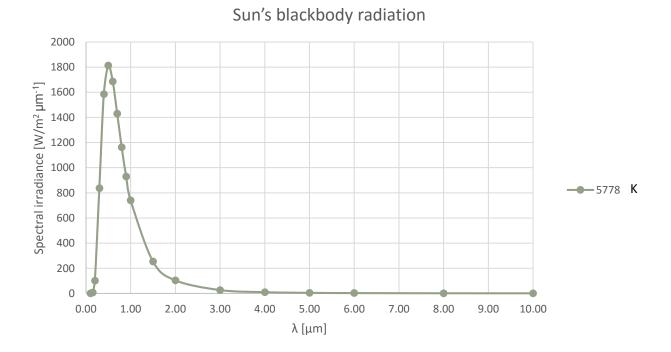
• The units of the Planck's law are per steradian. Therefore, multiplying the Spectral Irradiance by the steradians the sun occupies in Earth's sky we will get the Irradiance in W m<sup>-3</sup>





Receiver's efficiency:  $Q_{solar} = \eta_{optics} \cdot C \cdot A \cdot I$ 

Therefore, we get new values for the Irradiance





Receiver's efficiency: Q<sub>absorbed</sub>

 Q<sub>absorbed</sub> is the ammount of radiation absorbed by the receiver on top of the tower, and is defined as follows:

$$Q_{absorbed} = \alpha \cdot Q_{solar}$$

• Where  $\alpha$  =stands for absorptivity



Receiver's efficiency: Q<sub>lost</sub>

ullet  $Q_{lost}$  is the ammount of radiation lost due to emissions in the IR spectra of the receiver, it is defined as:

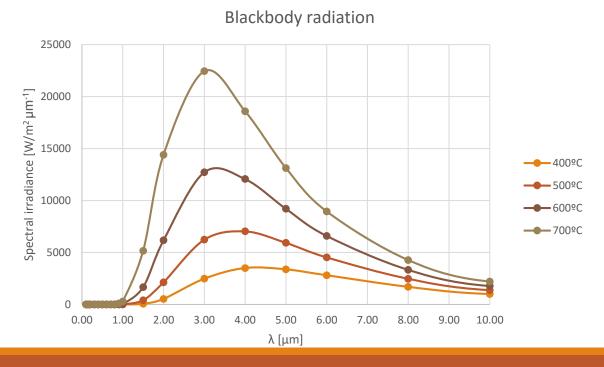
$$Q_{lost} = A \cdot \varepsilon \cdot \sigma \cdot T_H^4$$

• Where A = Receiver's area,  $\varepsilon$  is the surface's emissivity,  $\sigma$  stands for Steffan-Boltzmann constant and  $T_H$  = Receiver's surface temperature



Receiver's efficiency:  $Q_{lost} = A \cdot \epsilon \cdot \sigma \cdot T_H^4$ 

• If  $Q_{lost}$  is plotted per unit of area and assuming  $\epsilon = 1$  to show blackbody radiation of different temperatures





#### Total efficiency:

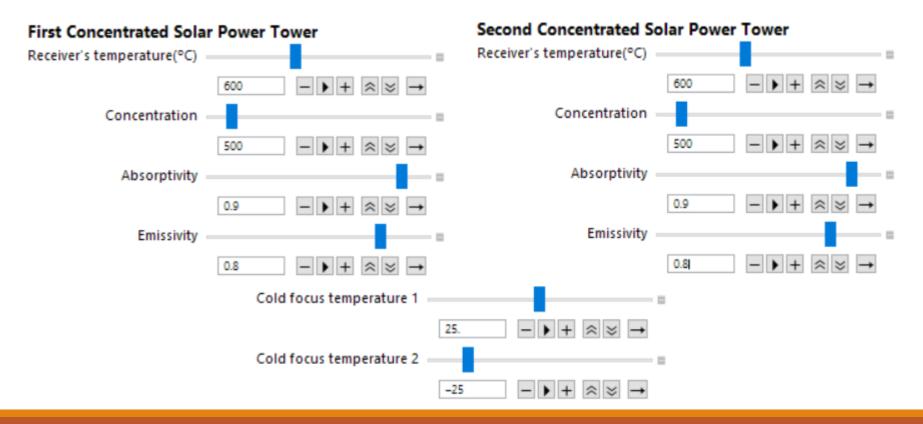
• The total efficiency is therefore defined as the product of receiver's and carnot's efficiencies.

$$\eta_{Receiver} = rac{Q_{absorbed} - Q_{lost}}{Q_{solar}}$$
 $\eta_{th} \leq 1 - rac{T_C}{T_H}$ 

• Then, 
$$\eta_{\text{total}} = \frac{Q_{absorbed} - Q_{lost}}{Q_{solar}} \left( 1 - \frac{T_c}{T_H} \right) = \frac{\alpha \cdot \text{C} \cdot \text{I} - \epsilon \cdot \sigma \cdot T_H^4}{\text{C} \cdot \text{I}} \left( 1 - \frac{T_c}{T_H} \right)$$



 Using Wolfram Mathematica, a computational model has been created to actively show how every variable affects efficiency

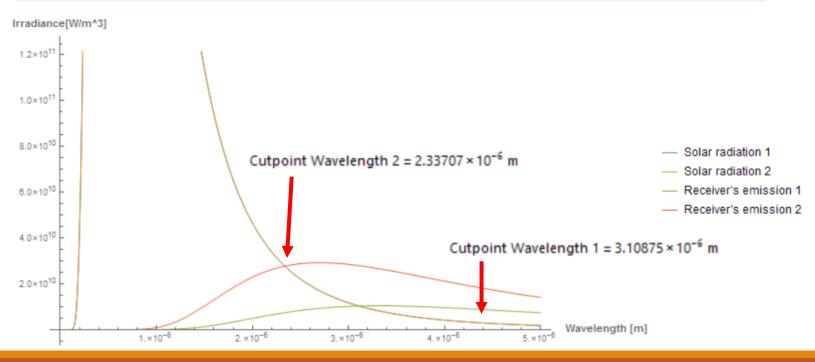




- In the following slides different situations will be shown with only one variable changed
- This way, how each one of the variables affect efficiency will be clearly detected
- Real-world experience of CSP reveals that there exists a 25-60 percent shortfall in projected production, so the following numbers and graph are theoretically correct but might not represent real CSP's numbers

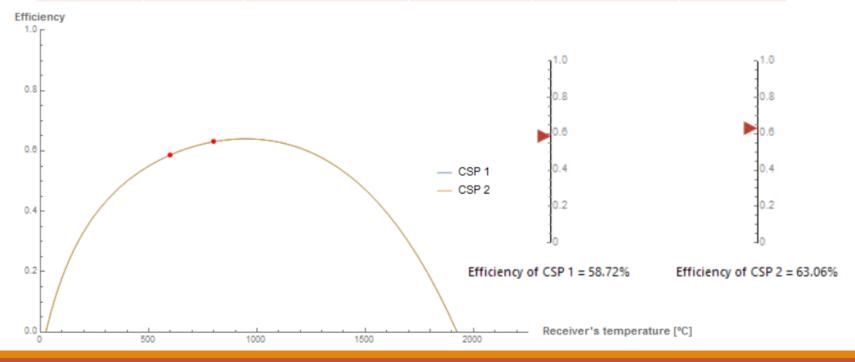


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (°C) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 800                   | 500           | 0.8        | 0.9          | 25                     |



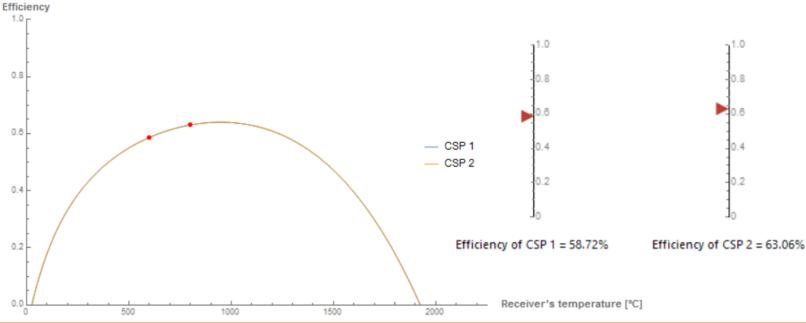


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 800                   | 500           | 0.8        | 0.9          | 25                     |



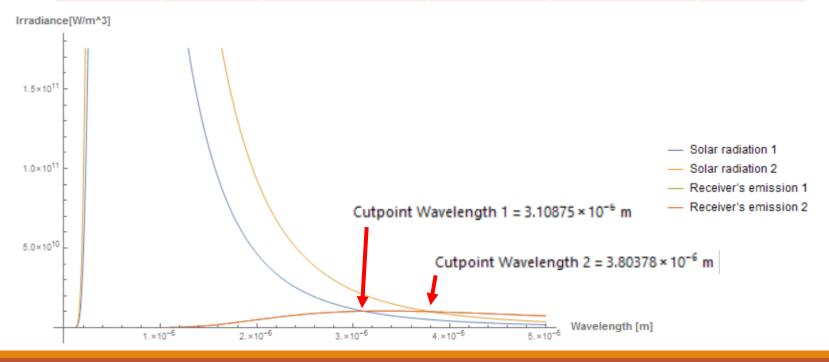


 The curve shows that there is a temperature for a maximum efficiency. In this case, raising temperature did not make the CSP move beyond that point and so overall efficiency goes up as IR emissions are less relevant than Carnot's efficiency and absorbed energy together.



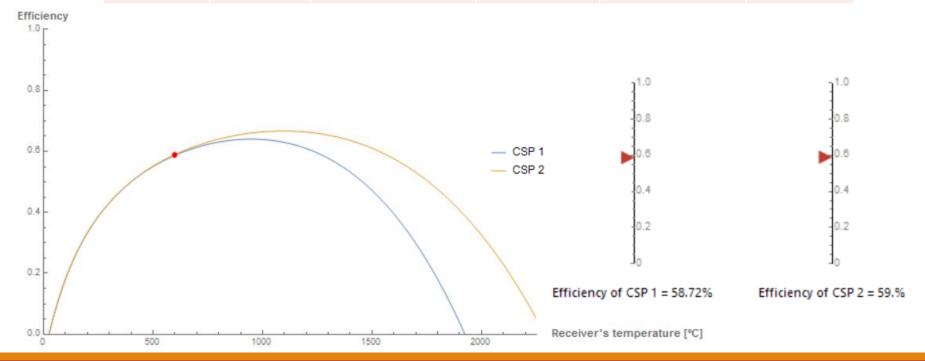


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 1000          | 0.8        | 0.9          | 25                     |



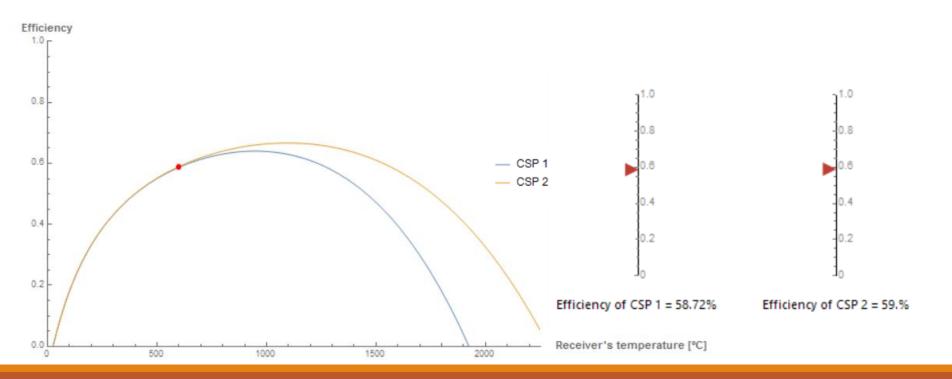


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (°C) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 1000          | 0.8        | 0.9          | 25                     |



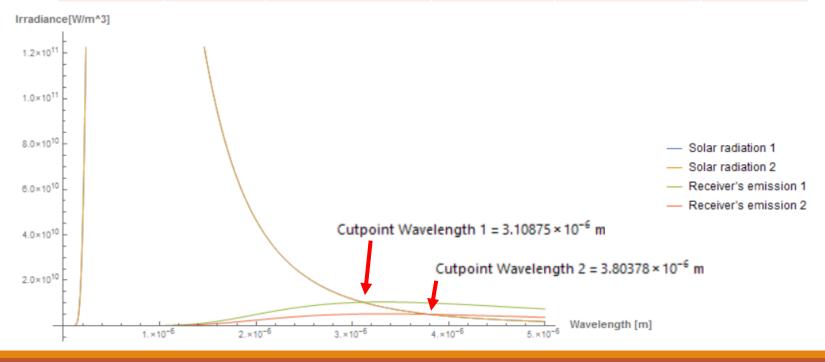


• Bigger concentrations move the efficiency curve upwards, as can be seen. The efficiency improvement is more noticeable at high temperatures (from the optimum temperature and on).



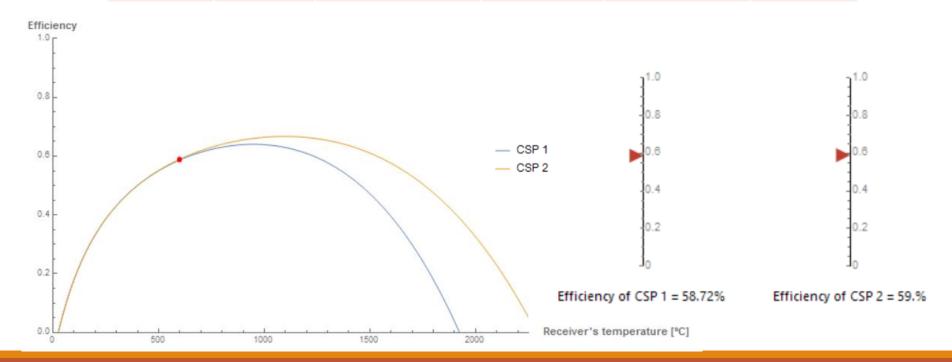


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 500           | 0.4        | 0.9          | 25                     |



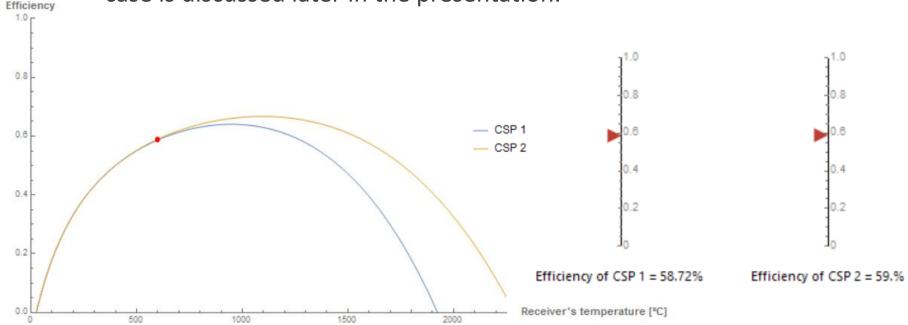


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 500           | 0.4        | 0.9          | 25                     |



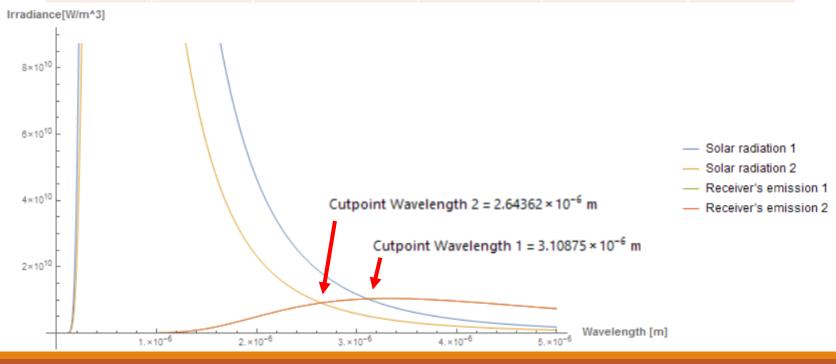


• Even if it is not easy to deduce from the formula at first sight, these plots and calculations show that the effect of multiplying the concentration by 2 and using a material with half the emissivity is exactly the same (at least in terms of efficiency and cut-point). This case is discussed later in the presentation.

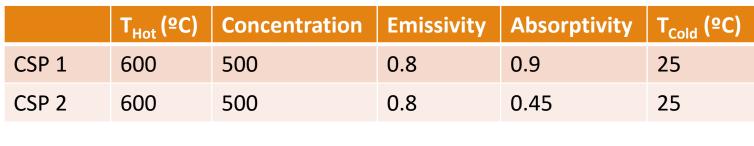


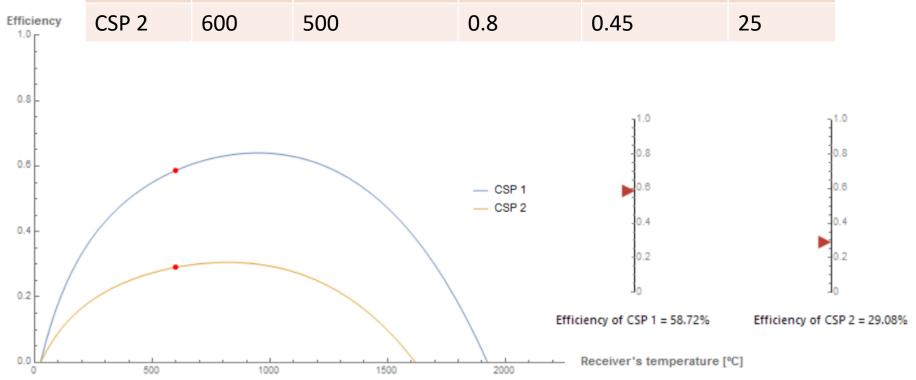


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 500           | 0.8        | 0.45         | 25                     |



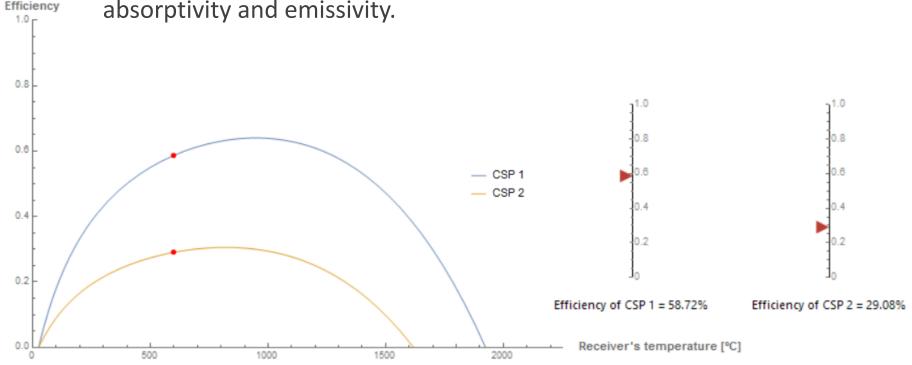






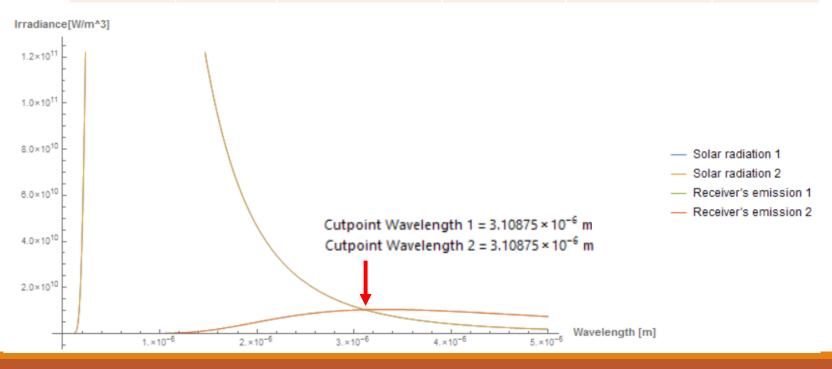


 Reducing the absorptivity has a bigger effect than reducing the emissivity, so keeping the rest of variables constant we would prefer a surface with high absorptivity and emissivity than one with low absorptivity and emissivity.



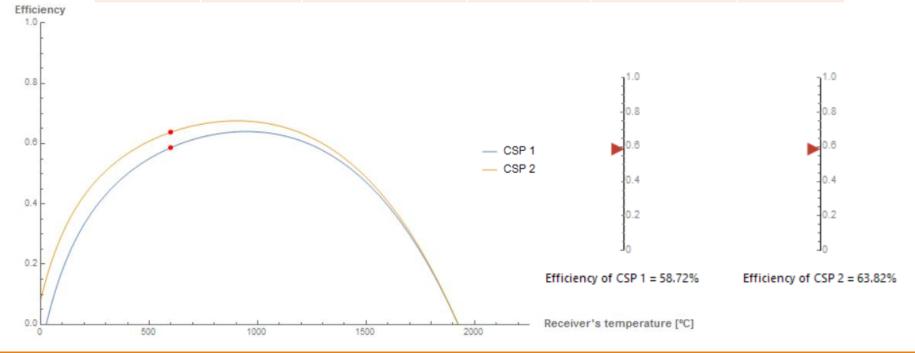


|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 500           | 0.8        | 0.9          | -25                    |





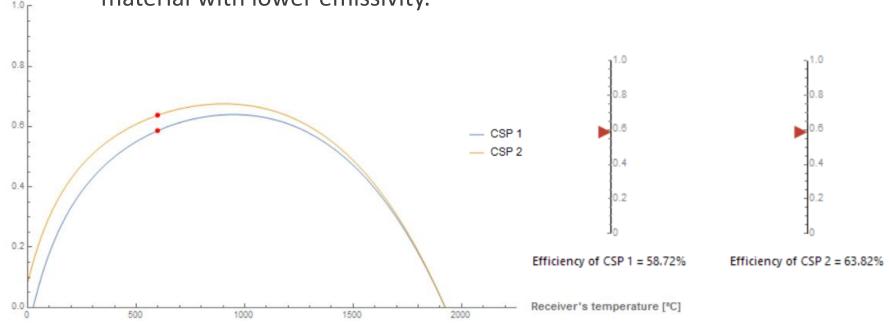
|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.8        | 0.9          | 25                     |
| CSP 2 | 600                   | 500           | 0.8        | 0.9          | -25                    |





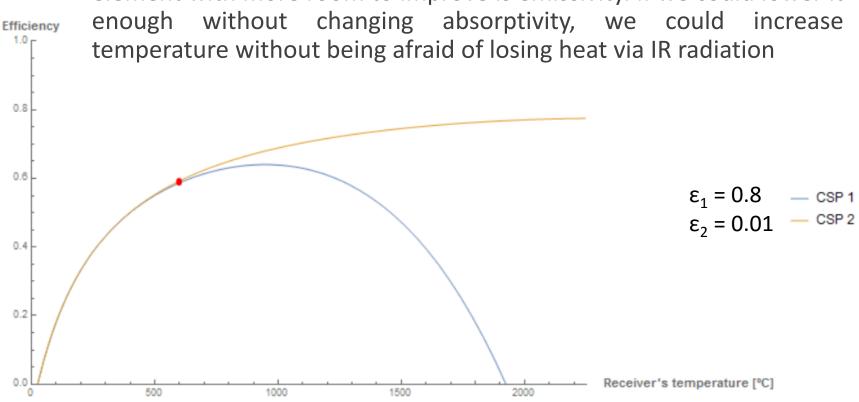
Efficiency

 Lowering the cold source's temperature seems to be as important (or even more) as lowering emissivities at low temperature CSP's. Specially because lowering the cold temperature is relatively easy (using liquid nitrogen for example) compared to developing a material with lower emissivity.



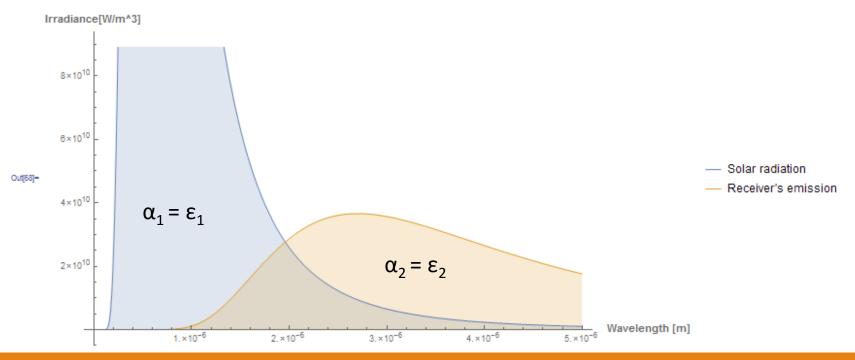


 As mentioned before, from the currently available technology, the element with more room to improve is emissivity. If we could lower it





However, every time absorption was being used up to this point, it
was applied to the solar spectrum only, and every time emissivity
was used, it was applied to the receiver's emission only, regardless of
their wavelength composition.

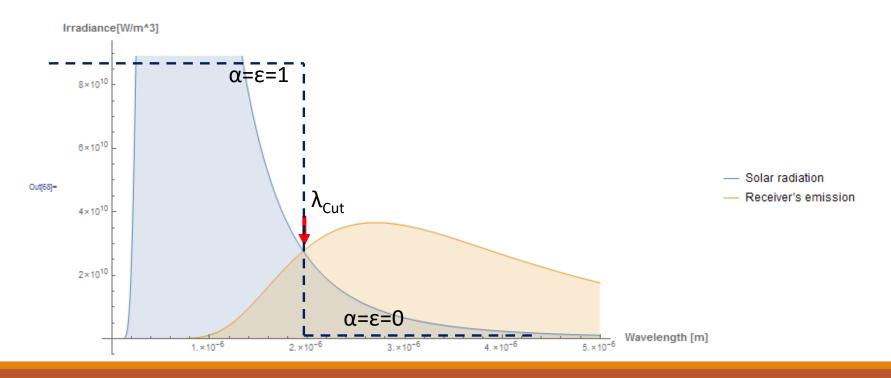




- That is how the ultimately perfect material would act. However, there is no way to design a material so that it has an absorption value for the whole solar spectrum and another value for the whole thermal emission.
- There are different ways, though, to design materials with different emissivity values for different wavelengths. So the ideal material would have an absorptivity value of 1 until the wavelength where the solar and the emission spectra cut, and a value of 0 from that point on.



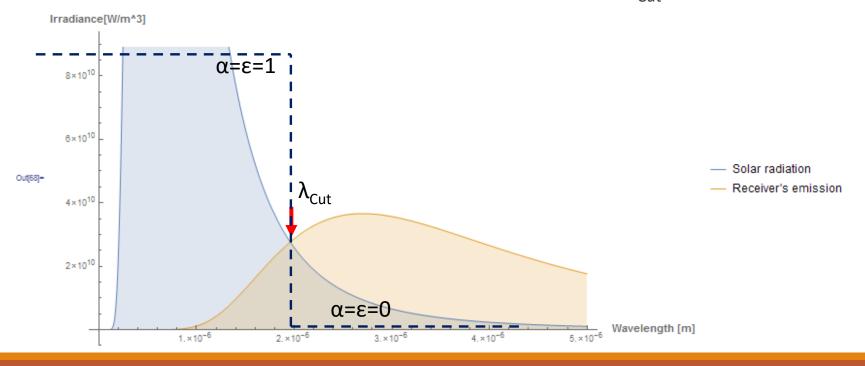
• In contrast to the previous model, blue shaded solar radiation from  $\lambda_{Cut}$  on would not be absorbed, and the orange shaded receiver's emission before  $\lambda_{Cut}$  would be irreversibly emitted.





#### Efficiency objective

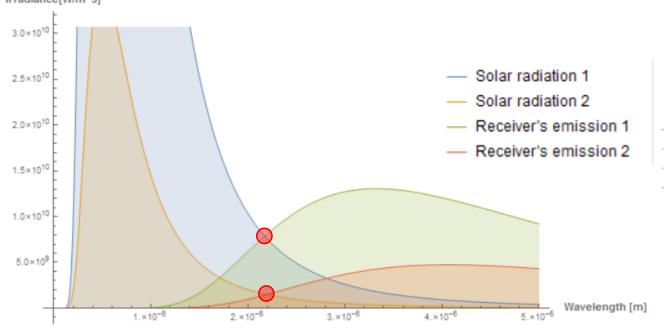
• This is in fact the big dilema of SPTs. The higher temperature the receiver works at, the more area that both curves have in common and so the more it is being not-absorbed from the sun (after  $\lambda_{\text{Cut}}$ ) and the more it is emitted from the receiver (before  $\lambda_{\text{Cut}}$ ).





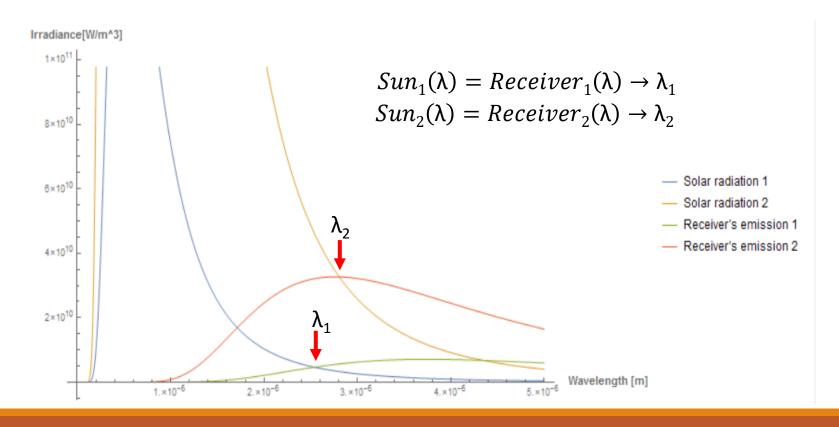
#### Efficiency objective

• The developed Mathematica model actively changes the Irradiance vs Wavelength graph, so two different models with different values of emissivity and absorptivity before and after the changing  $\lambda_{\text{Cut}}$  can be compared. The following slides are a short explanation of how it works.



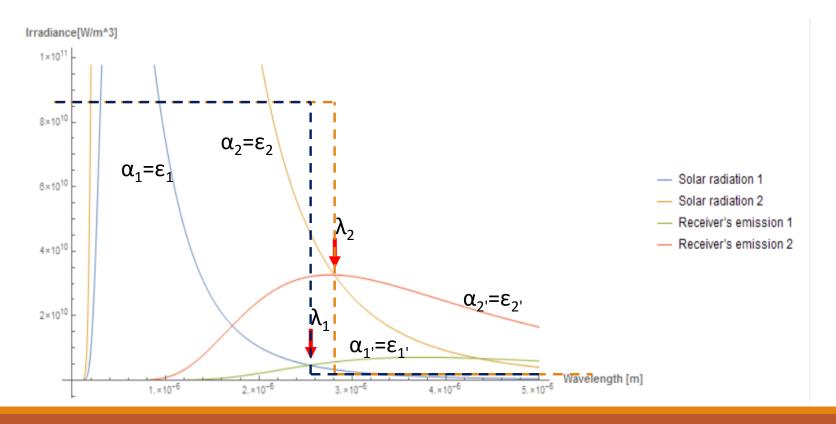


 Sun's blackbody radiation and receiver's emission curves are equaled and the cut-point is obtained





• Each couple of curves has two different emissivity values (that the user can choose), one for after and one for before each of the  $\lambda_{\text{Cut}}$ 





- Different values have been calculated for the variable values the user decided
- "Absorbed from the Sun" is the amount of radiation that the receiver absorbs, and is calculated as  $\int_0^{\lambda_{\text{Cut}}} Sun_1(\lambda) \cdot \alpha_1 d\lambda$  [W/m²]
- "Lost due to IR emission" is the amount of radiation the receiver emits in the wavelengths smaller than  $\lambda_{\text{Cut}}$ . It is calculated as  $\int_0^{\lambda_{\text{Cut}}} Receiver_1(\lambda) \cdot \alpha_1 \, d\lambda \, [\text{W/m}^2]$

|            | Absorbed from the sun [W/m2] | Lost due to IR emission [W/m2] | Total Absorbed [W/m2]   | Lost to absorbed ratio [%] |
|------------|------------------------------|--------------------------------|-------------------------|----------------------------|
| First CSP  | 131.43×10 <sup>3</sup>       | $3.41 \times 10^{3}$           | $128.023 \times 10^3$   | 2.596                      |
| Second CSP | 1.36×10 <sup>6</sup>         | 28.91×10 <sup>3</sup>          | 1.32706×10 <sup>6</sup> | 2.132                      |



- "Total absorbed" stands for the total amount of radiation the receiver absorbed minus that it emitted. It is obtaines as follows:  $\int_0^{\lambda_{\text{Cut}}} Sun_1(\lambda) \cdot \alpha_1 \, d\lambda \int_0^{\lambda_{\text{Cut}}} Receiver_1(\lambda) \cdot \alpha_1 \, d\lambda \, [\text{W/m}^2]$
- "Lost to absorbed Ratio" shows the ratio of the radiation lost to that being absorbed in the receiver. It is calculated with the following formula:  $\frac{\text{Lost due to IR emission}}{\text{Absorbed from the Sun}} \cdot 100 \, [\%]$

|            | Absorbed from the sun [W/m2] | Lost due to IR emission [W/m2] | Total Absorbed [W/m2]   | Lost to absorbed ratio [%] |
|------------|------------------------------|--------------------------------|-------------------------|----------------------------|
| First CSP  | 131.43×10 <sup>3</sup>       | $3.41 \times 10^{3}$           | $128.023 \times 10^3$   | 2.596                      |
| Second CSP | 1.36×10 <sup>6</sup>         | 28.91×10 <sup>3</sup>          | 1.32706×10 <sup>6</sup> | 2.132                      |



- "Not absorbed energy" is the amount of energy that is not being absorbed due to the emissivity change at  $\lambda_{\text{Cut}}$ . Its value is  $\int_{\lambda_{\text{Cut}}}^{\infty} Sun_1(\lambda) \cdot \epsilon_1 \, d\lambda \, [\text{W/m}^2]$
- "Not emitted energy" is the energy that it is not emitted due to the IR emission of the receiver. It is obtained as  $\int_{\lambda_{\text{Cut}}}^{\infty} Receiver_1(\lambda) \cdot \varepsilon_1 \, d\lambda$  [W/m²]

| Not absorbed energy [W/m2] | Not emitted energy [W/m2] | Absorption loses [%] | Emission savings [%] |
|----------------------------|---------------------------|----------------------|----------------------|
| $6.77174 \times 10^3$      | $62.4457 \times 10^3$     | 4.89974              | 94.8195              |
| $26.096 \times 10^3$       | $72.7247 \times 10^3$     | 1.8882               | 71.5579              |



• The "Absorption loses" define the ratio between the sun's notabsorbed energy from  $\lambda_{\text{Cut}}$  to the total solar energy (absorbed and not absorbed). It is calculated with the formula below:

$$\frac{\text{Not absorbed energy}}{\text{Absorbed from the Sun + Not absorbed energy}} \cdot 100 [\%]$$

 The "Emission savings" define the ratio between the not-emitted receiver's IR emission and the total IR emission of the receiver at the working temperature. It is calculated with the formula below:

$$\frac{\text{Not emitted energy}}{\text{Lost due to IR emission + Not emitted energy}} \cdot 100 [\%]$$

| Not absorbed energy [W/m2] | Not emitted energy [W/m2] | Absorption loses [%] | Emission savings [%] |
|----------------------------|---------------------------|----------------------|----------------------|
| $6.77174 \times 10^3$      | $62.4457 \times 10^3$     | 4.89974              | 94.8195              |
| $26.096 \times 10^3$       | $72.7247 \times 10^3$     | 1.8882               | 71.5579              |



 To show an example of its usefulness, the example discussed before were CSP1 has half the emissivity of CSP2 and CSP2 has twice the concentration that of CSP1 has been used. In the plots shown before, the cut-point and the efficiencies in both cases were exactly the same.

|       | T <sub>Hot</sub> (ºC) | Concentration | Emissivity | Absorptivity | T <sub>Cold</sub> (ºC) |
|-------|-----------------------|---------------|------------|--------------|------------------------|
| CSP 1 | 600                   | 500           | 0.4        | 0.9          | 25                     |
| CSP 2 | 600                   | 1000          | 0.8        | 0.9          | 25                     |

|            | Absorbed from the sun [W/m2] $$ | Lost due to IR emission [W/m2] | Total Absorbed [W/m2]  | Lost to absorbed ratio [%] |
|------------|---------------------------------|--------------------------------|------------------------|----------------------------|
| First CSP  | $614.98 \times 10^3$            | 9.08×10 <sup>3</sup>           | $605.9 \times 10^3$    | 1.476                      |
| Second CSP | 1.23×10 <sup>6</sup>            | 18.16×10 <sup>3</sup>          | 1.2118×10 <sup>6</sup> | 1.476                      |

| Not absorbed energy [W/m2] | Not emitted energy [W/m2] | Absorption loses [%] | Emission savings [%] |
|----------------------------|---------------------------|----------------------|----------------------|
| $6.94785 \times 10^3$      | $17.2628 \times 10^3$     | 1.11715              | 65.5311              |
| $13.8957 \times 10^3$      | $34.5257 \times 10^3$     | 1.11715              | 65.5311              |



• If both cases have the same efficiency, investing in doubling the amount of heliostats does not seem logical as it is much more expensive than using a material with half the emissivity. However, as proven by this model the amount of absorbed energy is way bigger in the case where the concentration is bigger, despite of having a worse emissivity. Therefore, the investment could have been justified.

|            | Absorbed from the sun [W/m2] | Lost due to IR emission [W/m2] | Total Absorbed [W/m2]  | Lost to absorbed ratio [%] |
|------------|------------------------------|--------------------------------|------------------------|----------------------------|
| First CSP  | $614.98 \times 10^3$         | $9.08 \times 10^{3}$           | $605.9 \times 10^3$    | 1.476                      |
| Second CSP | 1.23×10 <sup>6</sup>         | $18.16 \times 10^3$            | 1.2118×10 <sup>6</sup> | 1.476                      |

| Not absorbed energy [W/m2] | Not emitted energy [W/m2] | Absorption loses [%] | Emission savings [%] |
|----------------------------|---------------------------|----------------------|----------------------|
| $6.94785 \times 10^{3}$    | $17.2628 \times 10^3$     | 1.11715              | 65.5311              |
| 13.8957×10 <sup>3</sup>    | 34.5257×10 <sup>3</sup>   | 1.11715              | 65.5311              |

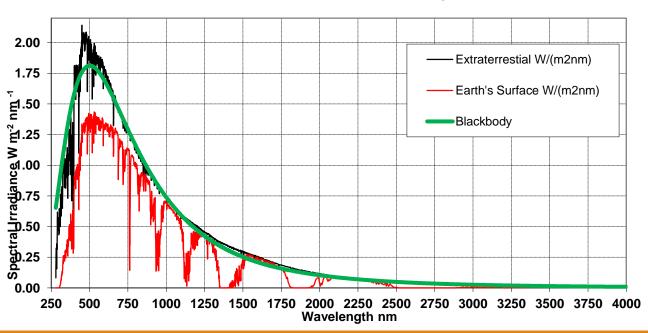


- However, even if sun's radiation has been assumed to be a blackbody radiation for simplicity's sake, it's actually not
- American Society for Testing and Materials (ASTM) and government research and development laboratories developed and defined a standard terrestrial solar spectral irradiance distribution
- "AM1.5", 1.5 atmosphere thickness, corresponds to a solar zenith angle of z=48.2°
- AM1.5 is useful to represent the overall yearly average for midlatitudes
- The specific value of 1.5 was selected in the 1970s for standardization purposes, based on an analysis of solar irradiance data in the United States



 The black line shows the extraterrestrial sun's radiation. The green line shows the blackbody radiation of a body at sun's surface temperature. The red line is the actual solar radiation at Earth's surface

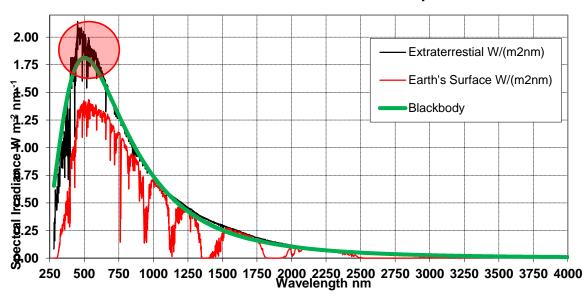
#### **ASTM G173-03 Reference Spectra**





 The main reason why this spectrum does not fit Planck's distribution is because in the mentioned distribution a single surface temperature is assumed. The sun is not a solid with a clearly defined outer face, but a ball of gas and plasma composed of several layers at different temperatures.

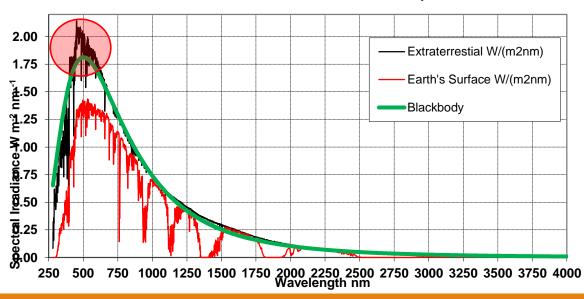






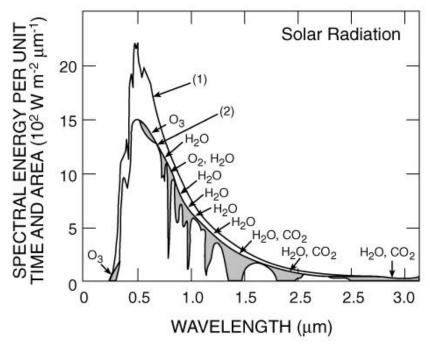
 In addition, the effective temperature of 5778 K is based on total radiative power, the area under the curve of the Planck distribution.
 If the spectrum of sunlight falls short of the 5778 K black body spectrum some wavelengths it must necessarily rise above the 5778 K black body spectrum at others

#### **ASTM G173-03 Reference Spectra**





 The discrepancies between the extraterrestrial radiation and the radiation that actually hits the Earth are explained with the absorption bands the elements in the atmosphere have



Source: www.paos.colorado.edu

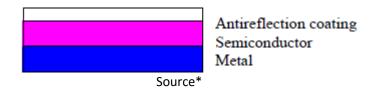
# Types of proposed solutions of Technology for CSP IR loss avoidance

- Multiple material structures have been proposed that meet the desired properties:
  - Semiconductor-metal tandems
  - Multilayer absorbers
  - Metal-dielectric composite coatings (Pyromark + nanofoam)
  - Textured surfaces

### Semiconductor-metal tandems



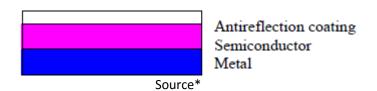
- Semiconductors with bandgaps from 0.5eV to 1.26eV that absorb short-wavelength radiation are used, and the underlaying metallic layer provides low emittance
- Antireflection coatings are needed because the useful semiconductors have high refractive indices
- The used semiconductors include Si(1.1eV), Ge(0.7eV), PbS(0.4eV)



### Semiconductor-metal tandems

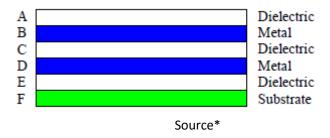


- The development of this type of materials were mainly developed in the 70's\*
- Even if good results were reported (absorptivity of 0.89 and emissivity of 0.05 at 500°C)\*, no recent research has been found in the field





- This type of absorber is formed by several layers of metallic and dielectric materials
- Overall, they all have the same structure:
  - D has high reflectance in the IR region and is slightly less reflective in the visible region
  - C reduces the visible reflectance
  - B further reduces the reflectance in the visible region
  - A increases the absorption in the visible region and broadens the high absorption region

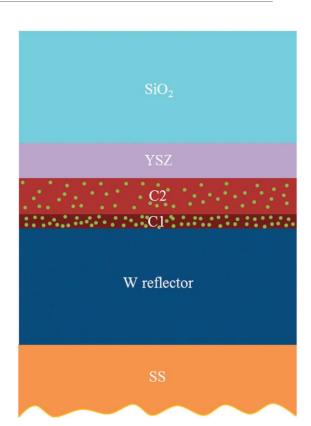




• The multilayer with the best results found to date is the developed in the next paper:

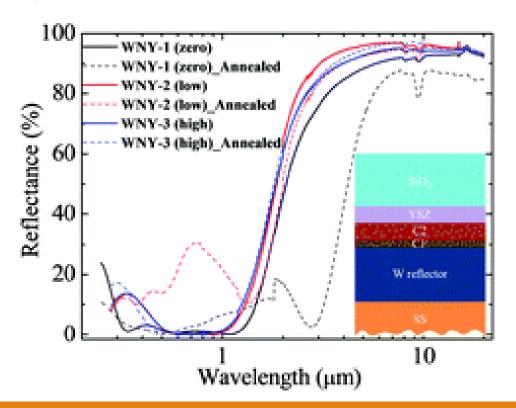
A high-performance spectrally-selective solar absorber based on a yttria-stabilized zirconia cermet with high-temperature stability, Feng Cao, Daniel Kraemer et al., The Royal Society of Chemistry, Energy Environ. Sci., 2015, 8, pp. 3040-3048 materials

• They developed a material with a solar absorptance of 0.91 and a total hemispherical emittance of 0.13 at 500°C





- The paper shows a reflectance vs wavelength graph
- The first step was to obtain the numerical data from the graph

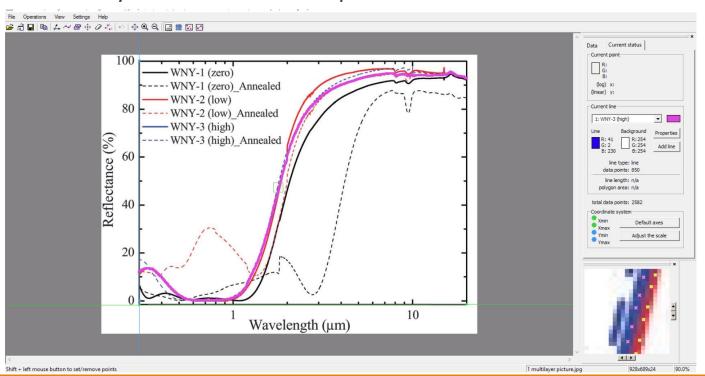




• In order to do so, *Graph Data* has been used



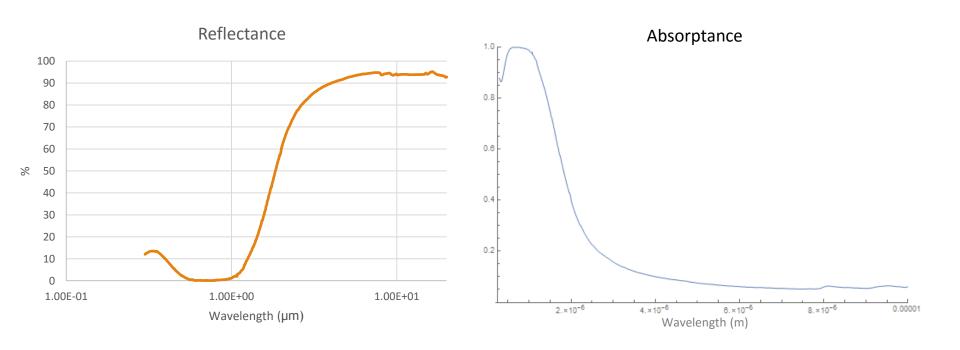
 Fixing the axis and plotting dots on top of the curve it calculates the dot's x and y coordinates and exports them to an excel sheet





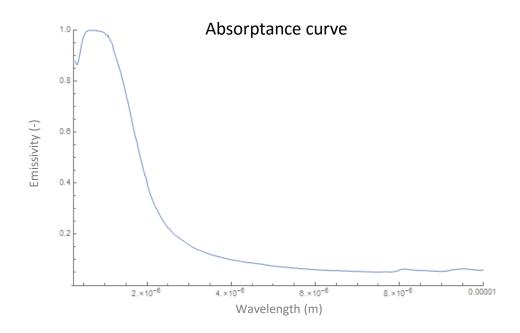
• From the excel sheet, the absorptance curve was obtained using:

Reflectance = 1- Absorptance



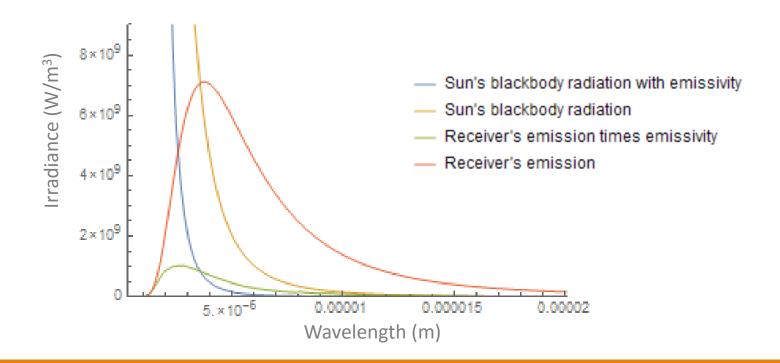


 Obtained curve is a discrete line (set of 800 points), and as it is not possible to multiply a function by a set of points, this set of point has been interpolated to a function with Wolfram Mathematica



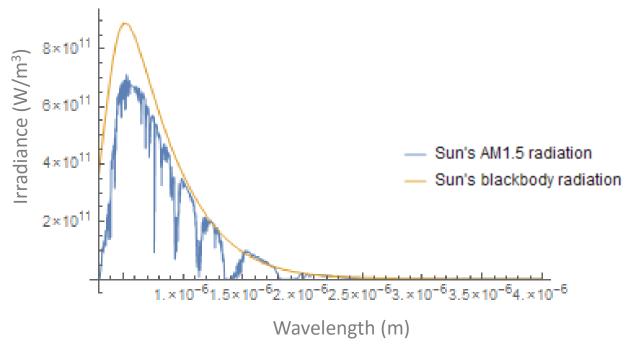


 Multiplying the interpolated emissivity by the sun's blackbody radiation and by the receiver's blackbody emission at 500°C (A concentration of 500 has been assumed)



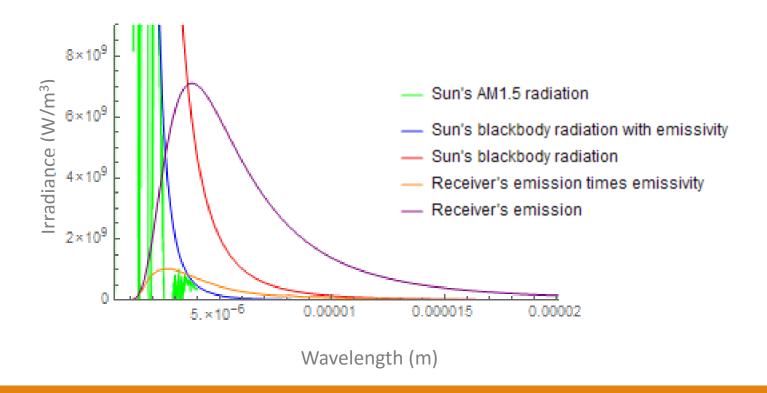


 Nevertheless, we mentioned that the true solar spectrum is not the blackbody radiation, so the data from the AM 1.5 spectrum has been imported to the program and the dots have been interpolated to obtain a function that would contain them all





 Plotting all the curves together (Sun's AM1.5 radiation has been multiplied by emissivity too),





 The areas of the graph have been calculated and the following numbers have been obtained:

| C=500          | With Multilayer (W/m2) | Without Multilayer (W/m2) |
|----------------|------------------------|---------------------------|
| Emitted        | 4,036.8                | 39,328.6                  |
| Absorbed       | 414,388                | 450,071                   |
| Total absorbed | 410,351                | 410,743                   |

- It is assumed that without the SSC the body acts as a blackbody
- Absorption improves 0.095 % without the Multilayer



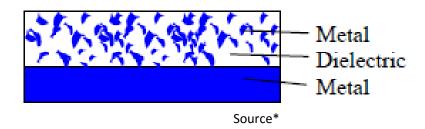
 Concentration plays an important role, as can be seen in the table below and comparing it to the previous example:

| C=100          | With Multilayer (W/m2) | Without Multilayer (W/m2) |
|----------------|------------------------|---------------------------|
| Emitted        | 4,036.8                | 39,328.6                  |
| Absorbed       | 82,877.6               | 90,014.2                  |
| Total absorbed | 78,840.8               | 50,685.6                  |

- It is assumed that without the SSC the body acts as a blackbody
- Absorption improves 55.549 % with the Multilayer



- This type of materials use a single or various layers of what are called cermets (a composite material formed of a ceramic matrix and metal particles)
- They usually are highly absorbing coatings in the solar region that are transparent in the IR, deposited onto a highly IR-reflective metal substrate\*
- The currently widely used Pyromark<sup>®</sup> 2500 series are paintings that fall into this type of coating





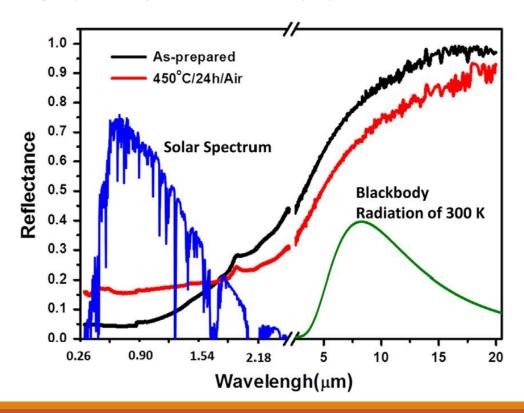
The best results found are in the following paper:

Xingli Wang, Xiaofend Wu, Long Yuan, Cuiping Zhou, Yanxiang Wang, Keke Huang, Shouhua Feng, *Solar selective absorbers with foamed nanostructure prepared by hydrothermal method on stainless steel*, Solar Energy Materials & Solar Cells 146 (2016), pp. 99–106

 They achieved a solar absorptance of 0.92 and a thermal emittance of 0.12

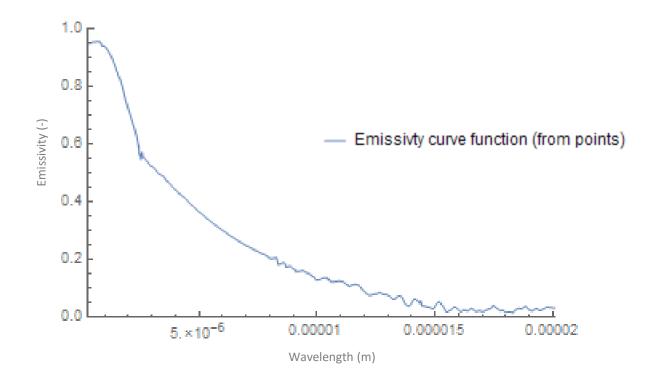


- However, they measured it at 300K
- This is the graph they show on the paper



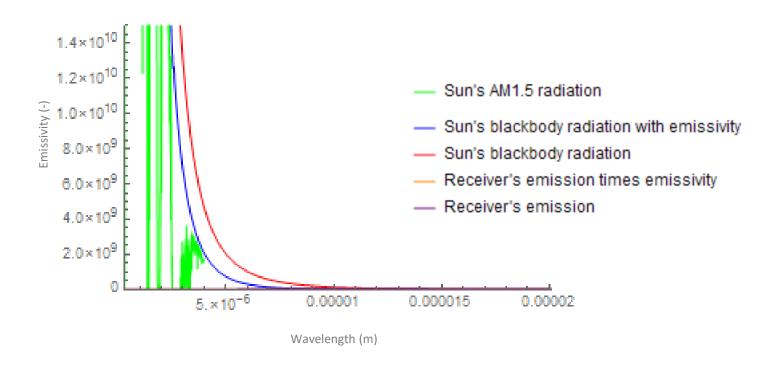


 Following the same process as with the multilayer absorbers, a function was created with the dots imported with the *Graph Data*





• The tested temperatures are so low that they do not even appear on the graph at this scale (Concentration of 500)





 The areas of the graph have been calculated and the following numbers have been obtained:

| C=500          | With Coating (W/m2) | Without Coating (W/m2) |
|----------------|---------------------|------------------------|
| Emitted        | 85.5                | 677.7                  |
| Absorbed       | 415,954             | 450,071                |
| Total absorbed | 415,869             | 449,394                |

- It is assumed that without the SSC the body acts as a blackbody
- Absorption improves 8.06 % without the Coating
- However there is no point on using this numbers as the materials has been tested at 300K



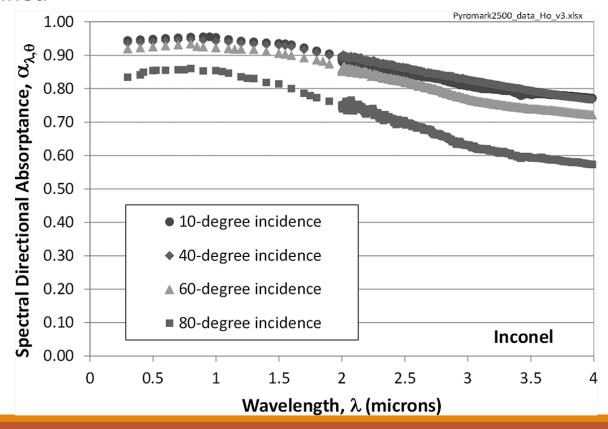
• The same procedure was followed to know what happens with the currently being used Pyromark® material. To do so, the following paper was studied:

Ho CK, Mahoney A, Ambrosini A, Bencomo M, Hall A, Lambert TN. Characterization of Pyromark 2500 Paint for High-Temperature Solar Receivers. ASME. *J. Sol. Energy Eng.* 2013;136(1):014502-014502-4. doi:10.1115/1.4024031.

 Pyromark® has an absorptance of up to 97% and an emittance of around 88% at 700°C

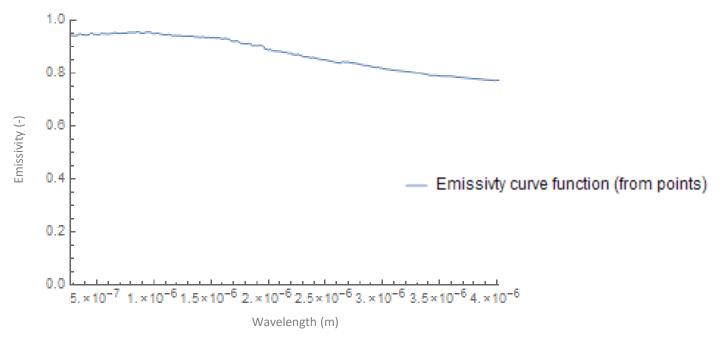


The study shows the following graph, from which data has been obtained



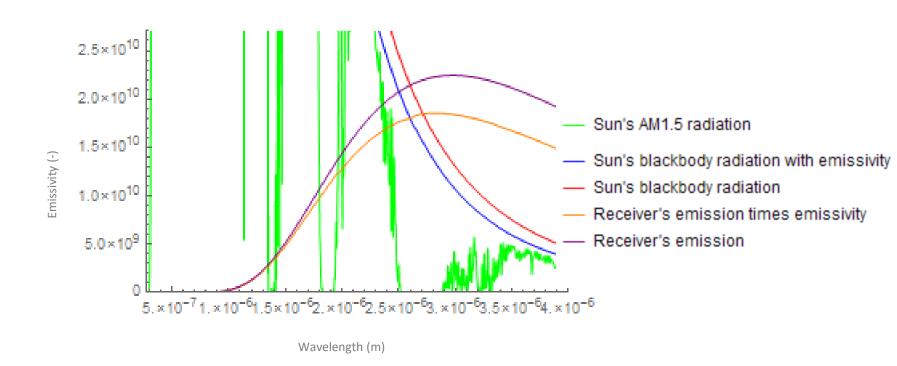


 The study does not show the whole wavelength spectrum, it only measures emissivity until 4 microns, which will limit the ability to compare it with the rest of the materials





This is the full graph up to 4 microns





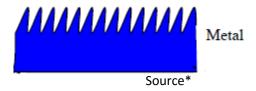
 The areas of the graph have been calculated and the following numbers have been obtained:

| C=500          | With Coating (W/m2) | Without Coating (W/m2) |
|----------------|---------------------|------------------------|
| Emitted        | 39,031              | 46,849.8               |
| Absorbed       | 424,529             | 450,071                |
| Total absorbed | 385,497             | 403,222                |

- It is assumed that without the SSC the body acts as a blackbody
- Absorption improves 4.6 % without the Coating
- However there is no point on using this numbers more than as an orientation way as the test has only ben performed up to 4 microns



- The aim of texturing the surface of a metal is to obtain spectral selectivity by the optical trapping of solar energy. The idea is to absorb solar energy while appearing highly reflective to thermal energy
- This technology is not very sensitive to severe environmental effects (oxidation, thermal shock...) that usually are catastrophic to the multilayer structures, but they are strongly affected by abrasion
- There are several fabrication techniques: Unidirectional solidification of eutectic alloys, lithography with X-rays, ion-exchange reactions between metals, vapor-liquid-solid mechanism, vapor deposition, oxidation of metals at high temperatures...





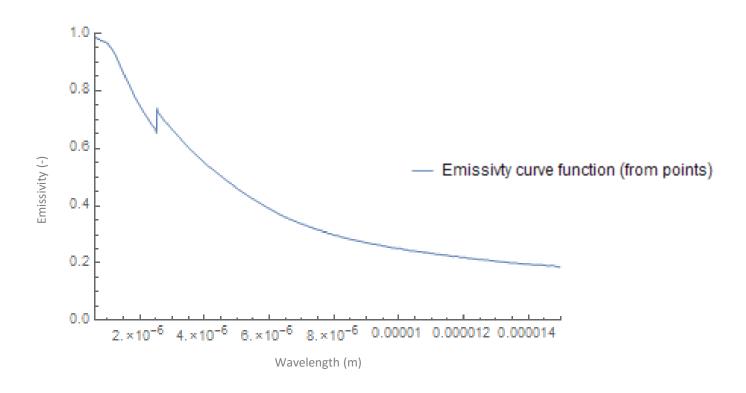
• Despite it was published in the 80's, the following paper shows some of the best results to the best of my knowledge:

G.L. Harding, M.R. Lake, Sputter etched metal solar selective absorbing surfaces for high temperature thermal collectors, Solar Energy Materials 5 (1981) pp. 445-464

• They achieved an absorptivity of 0.93 and an emissivity of 0.24 at 400°C

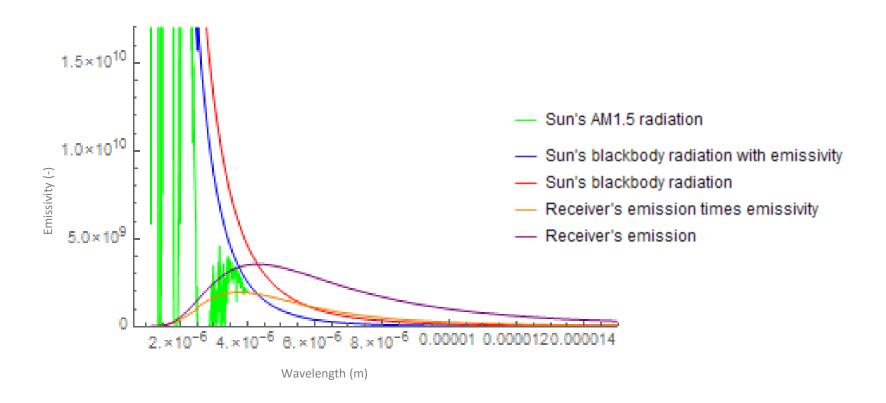


• Even if it is not recent, the following paper shows some of the best





• Even if it is not recet, the following paper shows some of the best





• The areas of the graph have been calculated and the following numbers have been obtained at 400°C:

| C=500          | With Textured Surface (W/m2) | Without Textured Surface (W/m2) |
|----------------|------------------------------|---------------------------------|
| Emitted        | 27,655                       | 22,308                          |
| Absorbed       | 284,858                      | 304,860                         |
| Total absorbed | 257,202                      | 282,552                         |

- It is assumed that without the SSC the body acts as a blackbody
- Absorption improves 9.85 % without the Textured Surface



- The calculation of the values have several sources of errors:
  - The reflectance curves provided by most of the studies show bidirectional reflectance, which can underestimate actual hemispherical emission in up to 30%
  - Getting set of points from a picture is not an accurate method of obtaining data
  - Solar AM 1.5 spetcrum's data ends at 4 microns
  - Calculated power values omit  $\eta_{optic}$  and  $\eta_{electric}$
  - Concentration and temperature have been assumed to be independent factors but further research must be done to find if there is any correlation



- Unless more information is found about the relationship between temperature and concentration, the Surface Selective Coatings (SSC) seem not to be that effective at high concentrations. The problem of the currently found SSCs is that the cost of lowering emissivity is reducing absorptivity. At high concentrations this is crucial because losing a 3% of solar absorptance, in absolute values can be more than what is saved reducing 90% of the emittance (specially at lower temperatures)
- Therefore, use of SSCs is justified at very high working temperatures and not that high concentrations (as long as they keep reducing absorptivity in order to lower emissivity)



- For temperatures below 1000°C, reducing emissivty can be as important as lowering the cold focus' temperature (specially if reducing emissivity implies reducing absorptivity)
- It is hard to compare different techniques and decide which one is best, as every research group has its own testing criteria and the spectra or temperature at which they test their materials is not the same
- Moreover, many research groups heat their materials to up to 600°C, but they test the emissivity back at room temperature which makes results incomparable



• For a coating to properly work at high temperatures it needs to undergo without losing its properties several heating and cooling cycles, and must be stable in air at the working temperatures. To the best of my knowledge, the materials that have been tested at higher temperatures were only stable in vacuum, and if they were stable in air either they lost their properties or were only tested for no more than 72 hours



# Metamaterials for Concentrated Solar Power applications

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